

Chapter

10

ELECTROCHEMISTRY

MULTIPLE CHOICE QUESTIONS

- Reduction always takes place: (LHR 05, 12)
 - At anode
 - At cathode
 - At both electrodes
 - does not occur at electrodes
- The electrolyte KOH is used in cell: (GRW 06)
 - Lead accumulator
 - Ni – Cd cell
 - Alkaline battery
 - Silver oxide battery
- In the reaction $2\text{Fe} + 3\text{Cl}_2 \rightarrow 2\text{FeCl}_3$: (BWP 10, GRW 09, LHR 06)
 - Fe is reduced
 - Fe is oxidized
 - Cl_2 is oxidized
 - None of these happens
- Oxidation state of hydrogen in CaH_2 is: (GRW 07)
 - +1
 - 1
 - +2
 - zero
- Cu metal can be purified in electrolytic cell by making the impure Cu as: (GRW 07)
 - Anode
 - Cathode
 - Anode and Cathode
 - Depends upon nature of solution
- Loss of electrons is called: (LHR 07)
 - Oxidation
 - Reduction
 - Hydration
 - Dehydration
- Fuel cells convert chemical energy into: (GRW 07)
 - Heat energy
 - Light energy
 - Electrical energy
 - Mechanical energy
- Electrolysis is used for: (LHR 08)
 - Electroplating
 - Manufacture of sodium metal
 - Manufacture of Al
 - All of these
- Nelson's cell and Down's cell are example of: (FSD 09, GRW 08)
 - Electrochemical cell
 - Galvanic cell
 - Electrolytic cell
 - None of these
- The oxidation number of Cr in $\text{K}_2\text{Cr}_2\text{O}_7$ is: (GRW 09, 10, LHR 10, 12, 13)
 - +14
 - +12
 - +6
 - +13
- The electrolyte used in fuel cell is: (LHR 09, DGK, GRW 10)
 - Aqueous NaCl
 - Molten NaCl
 - KOH
 - H_2SO_4

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