#### Assessment)

# **Chapter Test B**

# Chemistry of Life

Write the correct letter in the blank before each numbered term.

- \_\_\_\_\_ 1. covalent bond
  - 2. atom
- 3. reduction
- \_\_\_\_\_ 4. compound
- \_\_\_\_\_ 5. molecule
- 6. jonic bond
- \_\_\_\_\_ 7. oxidation
- 8, element

- a. involves transferred electrons
- b. loss of an electron
- c. substance that cannot be broken down
- d. involves shared electrons
- a substance composed of two or more elements
- f. gain of an electron
- g. simplest part of an element
- h. simplest part of a substance that can exist in a free state

Write the letter of the most correct answer in the blank.

- \_\_\_\_\_9. Elements tend to undergo chemical reactions that \_\_\_\_\_
  their stability.
  - a. increase

c. do not change

b. decrease

- d. disrupt
- \_\_\_\_\_10. Sodium chloride (table salt) is an example of a compound formed by
  - hydrogen bonding.
- ionic bonding.
- b. covalent bonding.

- d. electromagnetic bonding.
- \_\_\_\_11. When pure water dissociates, it forms equal numbers of
  - hydroxide ions and sodium ions.
  - b. hydronium ions and water ions.
  - c. hydroxide ions and water ions.
  - d. hydroxide ions and hydronium ions.
- \_\_\_\_\_12. Matter can change from a solid to a liquid by the addition of energy, which makes the particles in the matter
  - move faster.

c. dissociate.

b. move slower.

- d. increase their specific heat.
- 13. Charged particles that move around an atom's nucleus are
  - electrons.

c. ions.

b. protons.

d. neutrons.

# **Modern Biology Chapter Tests Answer Key**

**Yi-Tong Ma** 

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