

Chemistry – Unit 6 Worksheet 1

We have observed evidence that atoms of different elements are different, the most noticeable being in their naturally-occurring isotopes. Atoms of different particles exist in their naturally-occurring isotopes. However, when the elements are separated, isotopes of different elements are separated by their composition. The most common isotopes of an element are called "stable" isotopes. The most common isotopes of an element are called "stable" isotopes. We will now examine the patterns that exist for the stable isotopes of atoms. These atoms are called "stable" isotopes. We will now examine the patterns that exist for the stable isotopes of atoms.

1. Write the formula and draw the particle diagram for each compound. The table of atomic numbers is given.

Atomic number of element	12 elements 12 neutrons	12 elements 12 neutrons	12 elements 12 neutrons	12 elements 12 neutrons
	12	12	12	12

particle



diagram

Atomic number of element	12 elements 12 neutrons	12 elements 12 neutrons	12 elements 12 neutrons
	12	12	12

particle



diagram

Atomic number of element	12 elements 12 neutrons	12 elements 12 neutrons	12 elements 12 neutrons	12 elements 12 neutrons
	12	12	12	12

particle



diagram

Modeling Chemistry Unit 6 Test Answer Key

S Ashworth



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