

Stoichiometry Worksheet 2: Percent Yield

What aspects of a the percent yield calculation:

- Identify a balanced chemical equation of a reaction.
- Calculate the actual yield of a reaction (actual yield is what you actually get from a reaction, not what you think you will get).
- Identify the theoretical yield of a reaction (theoretical yield is what you would get if the reaction went to 100% completion).

1. Calculate the percent yield of a reaction. (Percent yield is the ratio of the actual yield to the theoretical yield, expressed as a percentage.)

$$\text{Percent Yield} = \frac{\text{Actual Yield}}{\text{Theoretical Yield}} \times 100\%$$

$$\text{Actual Yield} = \frac{\text{Percent Yield}}{100\%} \times \text{Theoretical Yield}$$

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2. Calculate the percent yield of a reaction. (Percent yield is the ratio of the actual yield to the theoretical yield, expressed as a percentage.)

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3. Calculate the percent yield of a reaction. (Percent yield is the ratio of the actual yield to the theoretical yield, expressed as a percentage.)

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4. Calculate the percent yield of a reaction. (Percent yield is the ratio of the actual yield to the theoretical yield, expressed as a percentage.)

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$$\text{Theoretical Yield} = \frac{\text{Actual Yield}}{\text{Percent Yield}} \times 100\%$$

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Kathleen Armour



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