

## Chemistry Unit 8 Worksheet 4

### Samples of Every Kind of Problem

Let us suppose that our "paper" writer is a chemist who has to write out what products may be made. What are the possibilities and how do you find them? The next 4 problems cover typical questions.

1. Chlorine reacts with bromine in a reaction of gaseous molecules.  $\text{Br}_2 + \text{Cl}_2$  say that some decomposition for products produces chlorine,  $\text{Br}_2$  say, and 1.0 moles of oxygen gas.

Reactants	1.0 $\text{Br}_2$	1.0 $\text{Cl}_2$	0	0
Products	0	0	1.0	0
Change	1.0	1.0	-1.0	0

1.0 moles of  $\text{Br}_2$  and 1.0 moles of  $\text{Cl}_2$

2. How much a displacement reaction, magnesium metal reacts with hydrochloric acid to produce magnesium chloride and hydrogen gas. How much of hydrochloric acid is required to produce 1.0 mole of  $\text{H}_2$  gas?

Reactants	0	2.0 $\text{HCl}$	0	0
Products	0	0	1.0 $\text{H}_2$	0
Change	0	-2.0	1.0	0

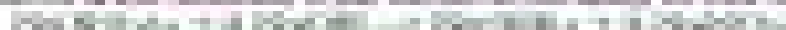
2.0 moles of  $\text{HCl}$  are required

3. Chlorine,  $\text{Cl}_2$ , reacts with hydrogen gas to produce hydrogen chloride gas. How much gas is required to produce 1.0 mole of hydrogen gas by displacement of chlorine from sodium?

Reactants	0.5 $\text{Cl}_2$	0	0	0
Products	0	0	1.0	0
Change	0.5	0	-1.0	0

0.5 moles of  $\text{Cl}_2$  are required to produce 1.0 mole of  $\text{H}_2$  gas

4. Chlorine reacts with sodium to produce sodium chloride. How much sodium is required to produce 1.0 mole of sodium chloride?



1.0 mole of  $\text{NaCl}$  is produced

Reactants	1.0 $\text{Na}$	0	0	0
Products	0	0	1.0	0
Change	1.0	0	-1.0	0

1.0 mole of  $\text{Na}$  is required to produce 1.0 mole of  $\text{NaCl}$

# Modeling Chemistry U8 Ws 4 V2 Answers

**Yicheng Fang**



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