

## Review Sheet for Chemistry\* First Semester Final

Refer to your class notes, worksheets, and the textbook to complete this review sheet. Study early so that you will have time to ask questions about what you don't understand.

### (\* The study of matter and the changes it undergoes)

**Matter: Anything that takes up space and has mass**

➤ Physical Changes and Chemical Changes

Define each. How can you tell the difference between the two?

Classify the following as physical or chemical changes:

- a. spoiling of milk \_\_\_\_\_
- b. bending wire \_\_\_\_\_
- c. cutting paper \_\_\_\_\_
- d. rusting of a nail \_\_\_\_\_

➤ Put the following into a graphic organizer/flowchart and define each:

- Matter
- Pure substances – elements and compounds
- Mixtures – homogeneous (solutions) and heterogeneous

Identify the following as pure substances, homogeneous mixtures or heterogeneous mixtures:

- a. copper \_\_\_\_\_
- b. sweetened tea \_\_\_\_\_
- c. sand and water \_\_\_\_\_
- d. calcium carbonate ( $\text{CaCO}_3$ ) \_\_\_\_\_

➤ Sketch particles in the three states of matter. How close are the particles and how much do they move?

Solid	Liquid	Gas

➤ Calculations using the Law of Conservation of Mass for Reactions



### Atom

➤ For this Carbon-14 isotope,  $^{14}_6\text{C}$

# Modeling Chemistry 1st Semester Final V2 3

**Mark Steyvers**



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