

## Unit 5 – Chemical Reactions

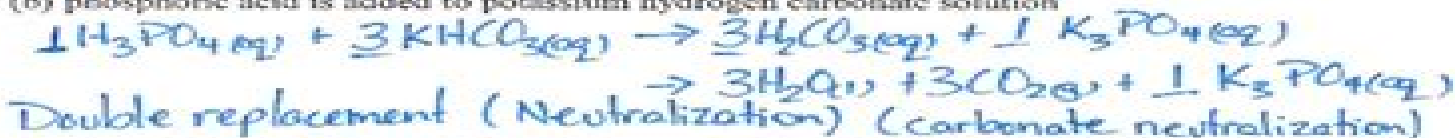
13 Write complete balanced chemical equations for the following. Classify each reaction as combination, decomposition, single replacement, double replacement, neutralization, or combustion. Don't forget your subscripts!

(a) potassium reacts with bromine



Combination

(b) phosphoric acid is added to potassium hydrogen carbonate solution



Double replacement (Neutralization) (carbonate neutralization)

(c) magnesium is added to nickel (II) nitrate solution



Single Replacement.

(d) hydrochloric acid is added to ammonium hydroxide.



Double Replacement (neutralization)

(e) nickel is added to magnesium chloride solution



(f) copper (II) sulphate pentahydrate is produced in a combination reaction



Combination

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**Lei Shi**



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## **Table of Contents Modeling Chemistry Unit 5 Test B V2 0 Key**

1. Understanding the eBook Modeling Chemistry Unit 5 Test B V2 0 Key
  - The Rise of Digital Reading Modeling Chemistry Unit 5 Test B V2 0 Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Modeling Chemistry Unit 5 Test B V2 0 Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Modeling Chemistry Unit 5 Test B V2 0 Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Modeling Chemistry Unit 5 Test B V2 0 Key
  - Personalized Recommendations
  - Modeling Chemistry Unit 5 Test B V2 0 Key User Reviews and Ratings
  - Modeling Chemistry Unit 5 Test B V2 0 Key and Bestseller Lists
5. Accessing Modeling Chemistry Unit 5 Test B V2 0 Key Free and Paid eBooks
  - Modeling Chemistry Unit 5 Test B V2 0 Key Public Domain eBooks
  - Modeling Chemistry Unit 5 Test B V2 0 Key eBook Subscription Services
  - Modeling Chemistry Unit 5 Test B V2 0 Key Budget-Friendly Options
6. Navigating Modeling Chemistry Unit 5 Test B V2 0 Key eBook Formats

- ePub, PDF, MOBI, and More
  - Modeling Chemistry Unit 5 Test B V2 0 Key Compatibility with Devices
  - Modeling Chemistry Unit 5 Test B V2 0 Key Enhanced eBook Features
7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Modeling Chemistry Unit 5 Test B V2 0 Key
    - Highlighting and Note-Taking Modeling Chemistry Unit 5 Test B V2 0 Key
    - Interactive Elements Modeling Chemistry Unit 5 Test B V2 0 Key
  8. Staying Engaged with Modeling Chemistry Unit 5 Test B V2 0 Key
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Modeling Chemistry Unit 5 Test B V2 0 Key
  9. Balancing eBooks and Physical Books Modeling Chemistry Unit 5 Test B V2 0 Key
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Modeling Chemistry Unit 5 Test B V2 0 Key
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Modeling Chemistry Unit 5 Test B V2 0 Key
    - Setting Reading Goals Modeling Chemistry Unit 5 Test B V2 0 Key
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Modeling Chemistry Unit 5 Test B V2 0 Key
    - Fact-Checking eBook Content of Modeling Chemistry Unit 5 Test B V2 0 Key
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
  14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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