

Name _____

Solution Stoichiometry Worksheet

Solve the following solutions Stoichiometry problems:

1. How many grams of silver chromate will precipitate when 150. mL of 0.500 M silver nitrate are added to 100. mL of 0.400 M potassium chromate?



$$\frac{0.150 \text{ L AgNO}_3}{1 \text{ L}} \times \frac{0.500 \text{ moles AgNO}_3}{1 \text{ L}} \times \frac{1 \text{ moles Ag}_2\text{CrO}_4}{2 \text{ moles AgNO}_3} \times \frac{331.74 \text{ g Ag}_2\text{CrO}_4}{1 \text{ moles Ag}_2\text{CrO}_4} = 12.4 \text{ g Ag}_2\text{CrO}_4$$

$$\frac{0.100 \text{ L K}_2\text{CrO}_4}{1 \text{ L}} \times \frac{0.400 \text{ moles K}_2\text{CrO}_4}{1 \text{ L}} \times \frac{1 \text{ moles Ag}_2\text{CrO}_4}{1 \text{ moles K}_2\text{CrO}_4} \times \frac{331.74 \text{ g Ag}_2\text{CrO}_4}{1 \text{ moles Ag}_2\text{CrO}_4} = 13.3 \text{ g Ag}_2\text{CrO}_4$$

2. How many mL of 0.280 M barium nitrate are required to precipitate as barium sulfate all the sulfate ions from 25.0 mL of 0.350 M aluminum sulfate? (93.8 mL barium nitrate)



$$\frac{0.0250 \text{ L Al}_2(\text{SO}_4)_3}{1 \text{ L}} \times \frac{0.350 \text{ moles Al}_2(\text{SO}_4)_3}{1 \text{ L}} \times \frac{3 \text{ moles Ba(NO}_3)_2}{1 \text{ moles Al}_2(\text{SO}_4)_3} \times \frac{1 \text{ L}}{0.280 \text{ moles Ba(NO}_3)_2} = 0.0938 \text{ L Ba(NO}_3)_2$$

3. 25.0 mL of 0.350 M NaOH are added to 45.0 mL of 0.125 M copper (II) sulfate. How many grams of copper (II) hydroxide will precipitate?



$$\frac{0.0250 \text{ L NaOH}}{1 \text{ L NaOH}} \times \frac{0.350 \text{ moles NaOH}}{1 \text{ L NaOH}} \times \frac{1 \text{ moles Cu(OH)}_2}{2 \text{ moles NaOH}} \times \frac{97.57 \text{ g Cu(OH)}_2}{1 \text{ mole Cu(OH)}_2} = 0.427 \text{ g Cu(OH)}_2$$

$$\frac{0.0450 \text{ L CuSO}_4}{1 \text{ L NaOH}} \times \frac{0.125 \text{ moles CuSO}_4}{1 \text{ L NaOH}} \times \frac{1 \text{ moles Cu(OH)}_2}{1 \text{ moles CuSO}_4} \times \frac{97.57 \text{ g Cu(OH)}_2}{1 \text{ mole Cu(OH)}_2} = 0.549 \text{ g Cu(OH)}_2$$

4. What volume of 0.415 M silver nitrate will be required to precipitate as silver bromide all the bromide ion in 35.0 mL of 0.128 M calcium bromide?



$$\frac{0.0350 \text{ L CaBr}_2}{1 \text{ L CaBr}_2} \times \frac{0.128 \text{ moles CaBr}_2}{1 \text{ L CaBr}_2} \times \frac{2 \text{ moles AgNO}_3}{1 \text{ moles CaBr}_2} \times \frac{1 \text{ L AgNO}_3}{0.415 \text{ mole AgNO}_3} = 0.0216 \text{ L AgNO}_3$$

5. What volume of 0.496 M HCl is required to neutralize 20.0 mL of 0.809 M sodium hydroxide?



$$\frac{0.0200 \text{ L NaOH}}{1 \text{ L NaOH}} \times \frac{0.809 \text{ mole NaOH}}{1 \text{ L NaOH}} \times \frac{1 \text{ mole HCl}}{1 \text{ mole NaOH}} \times \frac{1 \text{ L HCl}}{0.496 \text{ mole HCl}} = 0.0326 \text{ L HCl}$$

Mixed Stoichiometry Worksheet

Y Pai



Mixed Stoichiometry Worksheet:

Chemistry Carson-Dellosa Publishing, 2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series will be aligned to current science standards

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 This is a companion workbook for the 5 Steps to Surviving Chemistry book
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General Chemistry Workbook Daniel C. Tofan, 2010-07-28
 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions
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Collaborative Testing of NIOSH Atomic Absorption Method
 National Institute for Occupational Safety and Health. Division of Physical Sciences and Engineering, 1979

A Concrete Stoichiometry Unit for High School Chemistry Jennifer Louise Pakkala, 2006

Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998

Paper, 1999

ASME Technical Papers, 1999

Handbook on Material and Energy Balance Calculations in Material Processing, Includes CD-ROM Arthur E. Morris, H. Alan Fine, Gordon Harold Geiger, 2011-09-06
 This book approaches the subject of material and energy balances from two directions
 First it emphasizes the fundamental principles of the conservation of mass and energy and the consequences of these two principles
 Second it applies the techniques of computational chemistry to materials processing and introduces new software developed by the author especially for material and heat balances
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 The advance in computing power and software for the desktop computer has significantly changed the way engineers make computations and the biggest change comes from the computational approach used to solve problems
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Chemical Process Simulations using Aspen Hysys Khalid W. Hameed, 2025-07-16 An intuitive guide to using Aspen HYSYS for chemical petrochemical and petroleum industry process simulations including interactive process flow diagrams In Chemical Process Simulations using Aspen Hysys distinguished lecturer Dr Khalid W Hameed delivers an up to date and authoritative discussion of the simulation and design of chemical petrochemical and petroleum industry processes using Aspen HYSYS The book includes coverage of many chemical engineering topics including fluid flow reactors unit operation of heat and mass transfer oil refinery process and control systems Readers will also find highly interactive process flow diagrams for building and navigating through large simulations as well as A thorough introduction to the use of Aspen HYSYS for the chemical oil and petrochemical industries Skill development techniques for users of Aspen HYSYS and strategies for improving the accuracy of results Practical discussions of Dynamic State Simulation with explanations of how to install control systems for the process using flash separator gas processing and advanced process control such as ratio control cascade control and split range control Illustrative examples of Plant Wide Projects that demonstrate the ability of Aspen HYSYS to perform a full plant Perfect for research and development engineers in the fields of petrochemical chemical and petroleum engineering Chemical Process Simulations using Aspen HYSYS will also benefit researchers with an interest in the area

Working with Chemistry Donald J. Wink, Sharon Fetzer-Gislason, Julie Ellefson Kuehn, 2004-02-20 With this modular laboratory program students build skills using important chemical concepts and techniques to the point where they are able to design a solution to a scenario drawn from a professional environment The scenarios are drawn from the lives of people who work with chemistry every day ranging from field ecologists to chemical engineers and include many health professionals as well

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