

Mixed Stoichiometry Practice

Name _____

Date _____ Period _____

Write and/or balance the following equations (remember the diatomic elements and to cross-cross charges for ionic compounds!!!) Use the mole ratios from the balanced equations to solve the following stoichiometry problems. Use units and labels in all conversions, and round your answer to sig figs.

1. Potassium chlorate decomposes into potassium chloride and oxygen gas.

Balanced Equation:

2. How many moles of oxygen are produced when 3.0 moles of potassium chlorate decompose completely?

3. Butane (C_4H_{10}) undergoes combustion.

Balanced Equation: _____ $C_4H_{10}(l)$ + _____ $O_2(g)$ → _____ $CO_2(g)$ + _____ $H_2O(l)$

4. How many grams of CO_2 are produced when 60 g of O_2 are reacted with an excess of butane?

5. Water decomposes into hydrogen gas and oxygen gas by electrolysis.

Balanced Equation:

6. How many grams of hydrogen will be produced when 6.0 moles of oxygen are produced?

7. How many grams of water are required to produce 9.00 grams of hydrogen?

Mixed Stoichiometry Answers

D. G. Cacuci



Mixed Stoichiometry Answers:

Chemistry Carson-Dellosa Publishing, 2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series will be aligned to current science standards

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Fornari, C. Paorici, 1998-03-30 Proceedings of the 10th International Summer School on Crystal Growth Rimini Italy June 1998

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