

INTRODUCCIÓN

Muchas gracias por la adquisición de un motor Honda. Nos gustaría ayudarle a obtener los mejores resultados con su nuevo motor y a operarlo con seguridad. Este manual contiene información para ello; léalo detenidamente antes de poner en funcionamiento el motor. En caso de encontrarse con algún problema, o si tiene alguna pregunta sobre su motor, consulte a un concesionario de servicio Honda autorizado.

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Este manual debe considerarse como una parte permanente del motor y debe permanecer con el motor en caso de reventa.

Revise las instrucciones suministradas con el equipo que funcionará con este motor para encontrar información adicional sobre la puesta en marcha del motor, parada, operación, ajustes, o instrucciones especiales para el mantenimiento.

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MENSAJES DE SEGURIDAD

Su seguridad y la seguridad de los demás son muy importantes. Hemos incluido mensajes de seguridad importantes en este manual y en el motor. Lea detenidamente estos mensajes.

Un mensaje de seguridad le avisa sobre los peligros potenciales que podrían causarle lesiones a usted y a los demás. Cada mensaje de seguridad viene precedido por un símbolo de alerta de seguridad  y una de las tres palabras, PELIGRO, ADVERTENCIA o PRECAUCIÓN.

Estas palabras de indicación significan:

PELIGRO

Correrá el peligro de MUERTE o de HERIDAS GRAVES si no sigue las instrucciones.

ADVERTENCIA

Podrá correr el peligro de MUERTE o de HERIDAS GRAVES si no sigue las instrucciones.

PRECAUCIÓN

Podrá correr el peligro de HERIDAS si no sigue las instrucciones.

Cada mensaje le explica en qué consiste el peligro, lo que puede suceder, y lo que usted debe hacer para evitar las heridas o para reducirías.

MENSAJES PARA PREVENCIÓN DE DAÑOS

Encontrará también otros mensajes importantes que vienen precedidos por la palabra AVISO.

Esta palabra significa:

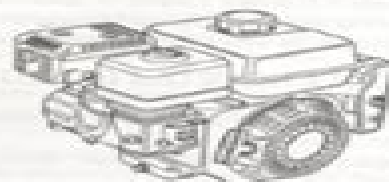
AVISO

Pueden producirse daños en el motor o en la propiedad de terceras personas si no sigue las instrucciones.

El propósito de estos mensajes es el de ayudar a evitar daños en el motor, en la propiedad de terceras personas, o en medio ambiente.

HONDA

MANUAL DEL PROPIETARIO GX120 · GX160 · GX200



ADVERTENCIA:

Los gases de escape de este producto contienen agentes químicos que, según el Estado de California, causan cáncer, defectos de nacimiento u otros efectos perjudiciales reproductivos.

ESPAÑOL

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Manual De Servio Honda Gx 160

David Baud



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web answer $\text{Na}^+ \text{aq}$ and $\text{SO}_4^{2-} \text{aq}$ identify the chemical equation the spectator ions $\text{KOH} \text{aq}$ HNO_3 $\text{KNO}_3 \text{aq}$ H_2O $\text{K}^+ \text{aq}$ $\text{OH}^- \text{aq}$ $\text{H}^+ \text{aq}$ $\text{NO}_3^- \text{aq}$ $\text{K}^+ \text{aq}$ $\text{NO}_3^- \text{aq}$ H_2O l answer $\text{K}^+ \text{aq}$ and $\text{NO}_3^- \text{aq}$ identify the chemical equation the spectator ions $\text{Pb(NO}_3)_2 \text{aq}$ $2\text{KI} \text{aq}$ $\text{PbI}_2 \text{s}$ $2\text{KNO}_3 \text{aq}$

4 3 precipitation reactions chemistry libretexts - Dec 03 2022

web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions gives the net ionic equation which shows only those species that participate in the chemical reaction $2\text{Ag}^+ \text{aq} + \text{Cr}_2\text{O}_7^{2-} \text{aq} \rightarrow 2\text{Cr}^{3+} \text{aq} + 7\text{H}_2\text{O} \text{l}$

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web chemical equation in which all dissolved ionic reactants and products including spectator ions are explicitly represented by formulas for their dissociated ions molecular equation chemical equation in which all reactants and products are

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web jan 29 2020 the ionic form of this reaction is $2 \text{Na}^+ \text{aq} + 2 \text{Cl}^- \text{aq} + \text{Cu}^{2+} \text{aq} + \text{SO}_4^{2-} \text{aq} \rightarrow 2 \text{Na}^+ \text{aq} + \text{SO}_4^{2-} \text{aq} + \text{CuCl}_2 \text{s}$ the sodium ions and sulfate ion are the spectator ions in this reaction they appear unchanged in both the product and reactant side of the equation

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web mar 11 2014 solution ionic equation $\text{H}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{Na}^+ \text{aq} + \text{HCO}_3^- \text{aq} \rightarrow \text{Na}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g}$ the spectator ions are Cl^- and Na^+
 net ionic equation $\text{H}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{Na}^+ \text{aq} + \text{HCO}_3^- \text{aq} \rightarrow \text{Na}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g}$
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web the six $\text{NO}_3^- \text{aq}$ ions and the six $\text{Na}^+ \text{aq}$ ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation $3\text{Ba}^{2+} \text{aq} + 2\text{PO}_4^{3-} \text{aq} \rightarrow \text{Ba}_3(\text{PO}_4)_2 \text{s}$

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web these spectator ions ions whose presence is required to maintain charge neutrality are neither chemically nor physically changed by the process and so they may be eliminated from the equation to yield a more succinct representation called a

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web video answer you know spectator ions are the ions present on the both sides of the equation indicate cancels out in the total ionic equations so you know in a case chemical reactions are described with chemical equations which symbolically present the starting materials end products

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web net ionic equations an equation for a reaction in solution showing only those particles that are directly involved in the chemical change spectator ions an ion that is not directly involved in a chemical reaction an ion that does not change oxidation number or composition during a reaction products

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water $\text{H}^+ \text{aq} + \text{OH}^- \text{aq} \rightarrow \text{H}_2\text{O} \text{l}$ the spectator ions in $\text{KOH} \text{aq}$ and $\text{HNO}_3 \text{aq}$ are $\text{K}^+ \text{aq}$ and $\text{NO}_3^- \text{aq}$ how to identify and list the

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web the ag aq and cl aq ions become agcl s but the na aq ions and the no 3 aq ions stay as na aq ions and no 3 aq ions these two ions are examples of spectator ions ions that do nothing in the overall course of a chemical reaction

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web aug 8 2022 a spectator ion is an ion that does not take part in the chemical reaction and is found in solution both before and after the reaction the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction

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web the new format requires students to write balanced chemical equations showing only the reacting substances excluding spectator ions for three different sets of reactants and to answer a short question requiring no calculator about each reaction the questions are intended to test the students understanding of the meaning of the reactions

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web jun 2 2023 the spectator ions are cut out from a reaction hence the net ionic equation for the above reaction would be 2cl aq cu 2 aq cucl 2 s 2cl aq cu 2 aq cucl 2 s these spectator ions are not actively involved in the reaction the existence of spectator ions affects the debye length

