Bubbles moving upward.

INTRODUCTION

LABORATORY SIMULATION



Lab Data

Verify your calculation. In your set-up, do your units cancel?

Did you report your data to the correct number of significant figures?

Mass of magnesium (g)		.015
Moles of magnesium (mol)	(0.00062
Temperature of water (°C)	N	23.0
Temperature of water (K)	10	296.15
Vapor pressure of water (mmHg)		21.1
Barometric pressure (mmHg)		781.0
Observations		

MY NOTES LAB DATA

Ideal Gas Law Lab Answers

Christian G. Meyer

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