

ICSE Class 10 Chemistry Question Paper Solution 2017

CHEMISTRY (SCIENCE PAPER-2)

SECTION I (40 Marks)

Attempt all questions from this Section

Question 1

- (a) Fill in the blanks from the choices given in brackets: [5]
- (i) The energy required to remove an electron from a neutral isolated gaseous atom and convert it into a positively charged gaseous ion is called _____.
(*electron affinity, ionisation potential, electronegativity*)
 - (ii) The compound that does not have a lone pair of electrons is _____.
(*water, ammonia, carbon tetra chloride*)
 - (iii) When a metallic oxide is dissolved in water, the solution formed has a high concentration of _____ ions. (*H⁺, H₃O⁺, OH⁻*)
 - (iv) Potassium sulphite on reacting with hydrochloric acid releases _____.
gas. (*Cl₂, SO₂, H₂S*)
 - (v) The compound formed when ethene reacts with Hydrogen is _____.
(*CH₄, C₂H₆, C₃H₈*)
- (b) Choose the **correct answer** from the options given below: [5]
- (i) A **chloride** which forms a precipitate that is soluble in excess of ammonium hydroxide, is:
 - 1. Calcium chloride
 - 2. Ferrous chloride
 - 3. Ferric chloride
 - 4. Copper chloride
 - (ii) If the molecular formula of an organic compound is **C₁₆H₁₈** it is:
 - 1. alkene

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