

Chapter 22

Hydrocarbon Compounds

22.1 Hydrocarbons

Organic Chemistry and Hydrocarbons

- Some thought that only living organisms could synthesize the carbon compounds found in living cells.
- German chemist **Friedrich Wöhler** was able to synthesize inorganic substances into synthetic organic compounds in 1828.
- Today, organic chemistry involves the study of all carbon compounds, regardless of origin.

Introduction to Hydrocarbons

Hydrocarbon compounds contain only hydrogen and carbon.

Alkananes are the simplest hydrocarbon compounds.

Alkanes contain four basic carbon structures in which each carbon atom is always bonded to four other atoms.

• The four simplest hydrocarbon compounds are methane, ethane, propane and butane, all having four single bonds and all saturated.



Alkanes are the simplest and most inert compounds in organic compounds. However, their inert or stable nature can make them difficult to synthesize.

→ They are also not burned by most simple fuels, such as wood. It was long known that carbon could burn into carbon dioxide.

The Properties of Hydrocarbons

Hydrocarbons are saturated molecules.

Hydrocarbons contain the simplest compounds that are saturated, i.e., they hydrocarbons with the lowest molecular weight for the given structure.

Alkanes are saturated hydrocarbons.

Alkanes are the simplest hydrocarbons, with four single bonds.

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Puja Mehta



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