

NAME: _____

HYDROCARBON NAMING WORKSHEET

- Naming (Given the structure be able to write the name)
 - Number longest carbon chain
 - Write the main group
 - Name and number the substituents (Alphabetically)
 - A prefix of (di, tri, tetra) is used when there are more than two of same substituent

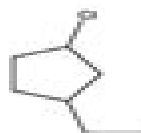
Naming Breakdown				
Numbers	two or more of same substituent	Substituents	Parent	endings
1	di	methyl	meth	-ane
2	tri	ethyl	eth	-ene
3	tetra	propyl	prop	-yne
4		butyl	but	benzene
5		fluoro	pent	
6		chloro	hex	
7		bromo	hept	
8		iodo	oct	
9		phenyl	non	
10		nitro	dec	
		sulfonic		
		cyclo		

1. Given the name draw the structure. Identify each 1°, 2°, 3°, or 4° carbon of an alkane.

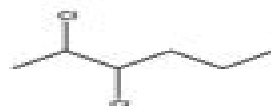
a) 3-methylheptane



b) 1-chloro-3-ethylcyclopentane



c) 2, 3-dichlorohexane



Hydrocarbon Study Guide Answer Sheet

Austin Roland



Hydrocarbon Study Guide Answer Sheet:

Hazmat Chemistry Study Guide (Second Edition) Jill Meryl Levy, 2011 **A Self-study Guide to the Principles of Organic Chemistry** Jiben Roy, 2013 A Self Study Guide to the Principles of Organic Chemistry Key Concepts Reaction Mechanisms and Practice Questions for the Beginner will help students new to organic chemistry grasp the key concepts of the subject quickly and easily as well as build a strong foundation for future study Starting with the definition of atom the author explains molecules electronic configuration bonding hydrocarbons polar reaction mechanisms stereochemistry reaction varieties organic spectroscopy aromaticity and aromatic reactions biomolecules organic polymers and a synthetic approach to organic compounds The over one hundred diagrams and charts contained in this volume will help students visualize the structures and bonds as they read the text and make the logic of organic chemistry clear and easily understood Each chapter ends with a list of frequently asked questions and answers followed by additional practice problems Answers are included in the Appendix **Study Guide with Selected Solutions** Karen Timberlake, 2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems Student Guide For Living Chemistry Barbara Ucko, 2012-12-02 Student Guide for Living Chemistry is a 23 chapter textbook guide that allows students to study and review on their own and test their understanding to help them prepare for examinations Every chapter begins with a list of objectives stating exactly the skills to develop in a particular unit Each objective corresponds to a section in the textbook Living Chemistry Three kinds of questions are provided for each objective to check the student's understanding namely short answer Study Questions multiple choice and fill in The answers for all questions are provided at the end of the chapter The opening chapters cover the SI units composition of matter chemical bonding compounds chemical change gases respiration and water The subsequent chapters deal with solutions acids bases salts nuclear and organic chemistry oxygen derivatives and hydrocarbons polymers and other organic derivatives This textbook also explores the chemistry of carbohydrates lipids proteins enzymes and energy and carbohydrate metabolism The remaining chapters discuss the chemistry of vitamins hormones body fluid drugs and poisons Undergraduate chemistry students will find this book invaluable **Chemistry, Student Study Guide** James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to

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Hydrocarbons in Marine Ecosystems and Organisms Douglas A. Wolfe,2013-09-24 Fate and Effects of Petroleum Hydrocarbons in Marine Organisms and Ecosystems is a collection of papers presented at the international symposium by the same title held at the Olympic Hotel in Seattle Washington on November 10 12 1976 This book is organized into three parts encompassing 46 chapters Part I deals with the inputs and physical transport processes influencing the distribution and composition of petroleum hydrocarbons in marine systems Part II discusses the bioaccumulation and metabolism of hydrocarbons by marine organisms Part III contains papers that tackle the biological and ecological effects of petroleum exposure in marine systems This book is of great value to marine and environmental scientists and researchers 200

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