

Study Guide for Chemistry EOC Assessment

Your test will consist of multiple-choice questions over the following topics.

1. Metric units for mass, length, volume
2. Metric to metric conversions (K H Da m/l/g D C M)
3. Know how to put numbers in scientific notation and take them out.
4. Significant Digit rules
5. Difference between chemical and physical properties and changes. There will be multiple questions on this.
6. Ways to tell if a chemical change (chemical reaction) is taking place.
7. Pure substances and mixtures
 - a. heterogeneous mixture
 - b. homogeneous mixture
8. Density & how it relates to floating and sinking
9. Use the equation for density
$$\text{Density} = \frac{\text{Mass}}{\text{volume}}$$

(g/ml) (g) (ml)
10. Know the phases of matter and the characteristics of each phase.
11. Law of Conservation of Matter
12. What is a mole in chemistry?
13. Convert grams to moles, and moles to grams
14. Define an atom
15. Structure of an atom and where each subatomic particle is located.
 - a. Nucleus (contains protons and neutrons)
 - b. Protons (positive)
 - c. Neutrons (neutral)
 - d. Electrons (negative)
16. Isotopes
17. Elements
18. If given atomic # and mass, find the number of protons and neutrons in an atom.
19. Find the number of valence electrons, and tell what it says about reactivity.
20. Periodic Table
 - a. Know about the families
 - b. Group vs. period
21. Know about the properties of metals and nonmetals.
 - a. Ductile, malleable, conductors
22. Ionic & Covalent Compounds

How do they form? (Transfer or gain electrons)

Characteristics of Each
23. Be able to name ionic compounds
24. Predict oxidation numbers
25. Chemical Equations:

Know the parts of an equation

5 main types of reactions-know about them and be able to identify them (Synthesis, Decomposition, Single Displacement, Double Displacement, Combustion)
26. Acids & Bases-

Macro and Microscopic properties of each

What ions do they form, how do they affect litmus, what is their pH range?

Identify examples of each- Mono, di, and polyprotic acids

Tell if they are strong or weak

Acid-base indicator

How To Study For The Tn Chemistry Eoc Exam

R Sanford



How To Study For The Tn Chemistry Eoc Exam:

Meeting the Challenges to Measurement in an Era of Accountability Henry Braun, 2016-01-29 Under pressure and support from the federal government states have increasingly turned to indicators based on student test scores to evaluate teachers and schools as well as students themselves The focus thus far has been on test scores in those subject areas where there is a sequence of consecutive tests such as in mathematics or English language arts with a focus on grades 4-8 Teachers in these subject areas however constitute less than thirty percent of the teacher workforce in a district Comparatively little has been written about the measurement of achievement in the other grades and subjects This volume seeks to remedy this imbalance by focusing on the assessment of student achievement in a broad range of grade levels and subject areas with particular attention to their use in the evaluation of teachers and schools in all It addresses traditional end of course tests as well as alternative measures such as portfolios exhibitions and student learning objectives In each case issues related to design and development psychometric considerations and validity challenges are covered from both a generic and a content specific perspective The NCME Applications of Educational Measurement and Assessment series includes edited volumes designed to inform research based applications of educational measurement and assessment Edited by leading experts these books are comprehensive and practical resources on the latest developments in the field The Open Access version of this book available at <http://www.taylorfrancis.com> has been made available under a Creative Commons Attribution Non Commercial No Derivatives 4.0 license

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Cracking the Virginia SOL Paul Foglino,Princeton Review (Firm),2000 The Princeton Review realizes that acing the Chemistry exam is very different from getting straight As in school The Princeton Review doesn t try to teach students everything there is to know about chemistry only the techniques they ll need to score higher on the exam There s a big difference In Cracking the Virginia SOL EOC Chemistry TPR will teach test takers how to think like the test makers and Score higher by mastering what test takers need to know with proven techniques Use word association to solve problems when unsure of the correct answers Ace the exam by understanding the logic of the test makers and applying it to solutions This book includes 2 full length simulated exams All of TPR s sample test questions are just like the ones test takers will see on the actual End of Course Chemistry exam and TPR fully explains every solution Contents Include The Mystery Exams About the SOL Exams How to Crack the Chemistry EOC II Chemistry Review Atomic Structure Anatomy of the Periodic Table Chemical Compounds Chemical Reactions Chemistry Math Gases Phase Changes Solutions Equilibrium Acids and Bases How Chemical Reactions Happen Laboratory III The Princeton Review Practice Tests Geographical Abstracts ,1986

Reverse Acronyms, Initialisms, & Abbreviations Dictionary ,1987 **Reverse Acronyms, Initialisms & Abbreviations Dictionary**. Mary Rose Bonk,Regie Carlton,Gale Research Inc,1997 Antarctic Bibliography ,1994 **TN Test Prep Bklt for Chemistry 2002** Holt Rinehart & Winston,2002-12-06 *Roadmap to the Virginia SOL* Princeton Review,2005 Roadmap to the Virginia SOL EOC Chemistryincludes strategies that are proven to enhance student

performance The experts at The Princeton Review provide content review of the crucial material most likely to appear on the test detailed lessons complete with test taking techniques for improving test scores 2 complete practice Virginia SOL EOC Chemistry tests

Texas EOC Chemistry - the All-In-one Guide Mary Amboji,2013-03-19 The complete guide for Texas EOC Chemistry has hundreds of guided and independent practice questions based on the categories tested on the EOC The dual coded questions help students gain mastery over scientific process skills Five category based tests are used to further hone your skills Calculator screen shots help you learn how to maximize your efficiency to solve complex problems There are two full length tests to prepare you for the actual exam This book can be used to prep students on chemistry

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