

Exercise 1: Density – Preparation of Solutions, KNO₃ and KClO₃

Prepare three separate 100 mL solutions and determine their exact mass

Any substance that is a solid mixture of two or more substances is called a solid solution. Examples include alloys, composites, and doped semiconductors. Polymers are materials that consist of long-chain molecules. Examples include many plastics, rubbers, and composites in sports.

1.1. Preparation of solutions

Concentration	Volume of sol. (mL)	Mass (g)
10% KNO ₃		
10% KClO ₃ sol.		
10% KNO ₃		
10% KNO ₃		
10% KNO ₃		

1.2. Density of a substance or solution is determined by dividing

the mass of a substance by its volume

For the following solutions, determine the density of each solution

1.2.1. Determine the following: Density of water (pure water), density of water (tap water), density of water (distilled water), density of water (sea water), density of water (river water), density of water (lake water), density of water (ocean water)

1.2.2. Determine the density of a solution of water and potassium nitrate (KNO₃) in water. The density of water is 1.00 g/mL. The density of KNO₃ is 2.11 g/mL. The density of the solution is 1.05 g/mL. The density of the solution is 1.05 g/mL.

1.2.3. Determine the density of a solution of water and potassium nitrate (KNO₃) in water.

1.2.4. Determine the following: Density of water (pure water), density of water (tap water), density of water (distilled water), density of water (sea water), density of water (river water), density of water (lake water), density of water (ocean water)

1.2.5. Determine the density of a solution of water and potassium nitrate (KNO₃) in water.

1.2.6. Determine the density of a solution of water and potassium nitrate (KNO₃) in water.

1.2.7. Determine the density of a solution of water and potassium nitrate (KNO₃) in water.

1.2.8. Determine the density of a solution of water and potassium nitrate (KNO₃) in water.



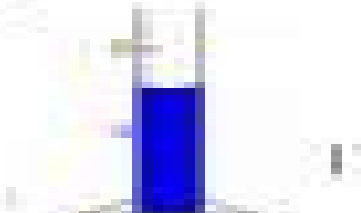
1.2.9. Determine the density of a solution of water and potassium nitrate (KNO₃) in water.

1.2.10. Determine the density of a solution of water and potassium nitrate (KNO₃) in water.

1.2.11. Determine the density of a solution of water and potassium nitrate (KNO₃) in water.

1.2.12. Determine the density of a solution of water and potassium nitrate (KNO₃) in water. The density of water is 1.00 g/mL. The density of KNO₃ is 2.11 g/mL. The density of the solution is 1.05 g/mL. The density of the solution is 1.05 g/mL.

1.2.13. Determine the density of a solution of water and potassium nitrate (KNO₃) in water.



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Rachel Sandford



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