

Q.41 The vapour pressure of pure liquid B at temperature T_0 is 0.5 atm. The partial pressure of B in the vapour phase that is in equilibrium with the liquid solution consisting of 30 mol% A and 70 mol% B at temperature T_0 is (assume both liquid and vapour phases behave ideally)

- (A) 0.35 atm (B) 0.50 atm (C) 0.70 atm (D) 1.00 atm

Q.42 During low temperature plastic deformation of an under-aged precipitation hardened alloy, dislocations

- (A) climb to completely avoid the precipitate
(B) loop around the precipitate
(C) cross-slip to completely avoid the precipitate
(D) cut through the precipitate

Q.43 According to Hume-Rothery rules, extensive solid solubility between elements X and Y is promoted by the two factors in the following list :

- P. Same crystal structure of X and Y
Q. Large atomic size difference ($> 20\%$) between X and Y
R. Same valence of X and Y
S. Large difference in melting points of X and Y

- (A) P, Q (B) P, R (C) Q, S (D) P, S

Q.44 At constant temperature and pressure, two phases α and β will be in equilibrium when

- (A) chemical potential of each component is the same in α and β
(B) partial molar free energy of each component is NOT the same in α and β
(C) Gibbs free energy of mixing is minimum
(D) enthalpy of mixing is zero

Q.45 The stress applied on a material is

$$\sigma_y = \begin{bmatrix} 21 & 0 & 0 \\ 0 & 21 & 0 \\ 0 & 0 & 21 \end{bmatrix} \text{ MPa.}$$

The maximum shear stress experienced by it is

- (A) 0 MPa (B) 10.5 MPa (C) 21 MPa (D) 63 MPa

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