

Student Name: (Last, First, Middle Initial) _____, Section: _____, Professor's Name: _____, Date: _____

1. Dry ice (solid carbon dioxide) is often called "snakes." A 1.00 g piece of dry ice is placed in a 1.00 L container that is evacuated at 100. °C. What is the pressure in the container after all of the dry ice has sublimated?

$$PV = nRT$$

$$\frac{1.00 \text{ g CO}_2}{44.01 \text{ g/mol}} \times \frac{1 \text{ mole CO}_2}{22.4 \text{ L @ 0°C}} = 0.0100 \text{ moles CO}_2$$

$$P = ?$$

$$V = 1.00 \text{ L (1 significant figure)}$$

$$n = 0.0100 \text{ moles CO}_2 \text{ (significant figures)}$$

$$R = 0.0821 \text{ L} \cdot \text{atm}/\text{mol} \cdot \text{K} \text{ (constant; 4 significant figures)}$$

$$T = 100. \text{ °C} = 373.15 \text{ K (1 significant figure)}$$

$$P = \frac{nRT}{V} = \frac{(0.0100 \text{ moles}) (0.0821 \text{ L} \cdot \text{atm}/\text{mol} \cdot \text{K}) (373.15 \text{ K})}{1.00 \text{ L}} = 0.307 \text{ atm}$$

20 grams of substance B, from the same source

- a) 1.00 atm
- b) 1.00 atm·hr
- c) 0.307 atm

2. A sample of chlorine gas is collected over a 0.025 L bottle at atmospheric temperature and pressure. How many moles of chlorine gas are in the container? How many grams?

$$0.025 \text{ L sample} = 0.025 \text{ L}$$

$$\text{Molar Mass of Chlorine (molecular, or as a diatomic)} = 70.9 \text{ g/mol, grams}$$

How many moles

$$\frac{0.025 \text{ L}}{22.4 \text{ L/mol}} = \frac{1 \text{ mole CO}_2}{22.4 \text{ L @ 0°C}} = 0.0011 \text{ moles CO}_2$$

$$\frac{0.025 \text{ L}}{22.4 \text{ L/mol}} \times \frac{1 \text{ mole CO}_2}{22.4 \text{ L @ 0°C}} = \frac{70.9 \text{ g/mol CO}_2}{1 \text{ mole CO}_2} = 0.077 \text{ g CO}_2$$

$$0.025 \text{ L (one (1) value) gas law, } PV = nRT$$

$$P = 1 \text{ atm (constant; significant figures)}$$

$$V = 0.025 \text{ L (1 significant figure)}$$

$$R = 0.0821 \text{ L} \cdot \text{atm}/\text{mol} \cdot \text{K} \text{ (constant; 4 significant figures)}$$

$$T = 273 \text{ K (constant; significant figures)}$$

Gas Laws Extra Practice

LM Reder



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web das buch mit den verbtabelle ist ein sinnvolles nachschlagewerk und für die praktische anwendung der italienischen sprache sehr hilfreich es erleichtert das benutzen von verben in unterschiedlichen zeiten