

## WS Gas Laws - KEY

### Boyle's Law

$$P_1 \times V_1 = P_2 \times V_2 \quad \text{At constant temperature}$$

### Charles's Law

$$\frac{V_1}{T_1} = \frac{V_2}{T_2} \quad \begin{array}{l} \text{At constant pressure} \\ \text{Temperature in Kelvin} \\ \text{ } ^\circ\text{C} + 273 = \text{K} \end{array}$$

### Lussac's Law

$$\frac{P_1}{T_1} = \frac{P_2}{T_2} \quad \begin{array}{l} \text{At constant volume} \\ \text{Temperature in Kelvin} \end{array}$$

### Combined Gas Law

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

STP: 273 K & 1.00 atm

### Concepts: (circle one choice)

- If the volume of a container of gas is reduced, what will happen to the pressure inside the container?  
increase or decrease
- What happens to the volume of a gas when it is cooled to a lower temperature?  
increase or decrease
- What happens to the pressure in a rigid container when it is slowly warmed?  
increase or decrease

### Sample Calculations:

(Remember: ALL temperatures MUST be in K, NOT °C)

- A sample of gas at 240 K and 670 mmHg occupies a 0.128 L volume. What volume will the gas occupy at 198 K if the pressure remains constant?

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad \frac{(0.128 \text{ L})}{(240 \text{ K})} = \frac{V_2}{(198 \text{ K})} \quad (0.128)(198) = V_2 = \boxed{0.106 \text{ L}}$$

- A sample of gas is in a steel container at  $-75.0^\circ\text{C}$  and 1.48 atm. At what temperature will the sample have a pressure of 7.35 atm?

$$\text{K} = -75 + 273 = 198 \text{ K}$$

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad \frac{(1.48 \text{ atm})}{(198 \text{ K})} = \frac{(7.35 \text{ atm})}{T_2} \quad T_2(1.48) = (7.35)(198) \quad T_2 = \frac{(7.35)(198)}{(1.48)} = \boxed{983 \text{ K}}$$

- In an airplane, a gas sample occurs at a volume of 1.50 L at 760 mmHg. Suppose, while flying, the airplane loses pressure and the volume of the gas increases to 11.40 L. What is the pressure in the airplane if the temperature is constant?

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad (760 \text{ mmHg})(1.50 \text{ L}) = P_2(11.40 \text{ L}) \quad (760)(1.50) = P_2 = \boxed{100 \text{ mmHg}}$$

# Gas Law Simulation Lab Answer Key

**Yicheng Fang**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## **Gas Law Simulation Lab Answer Key:**

The Top Books of the Year Gas Law Simulation Lab Answer Key The year 2023 has witnessed a noteworthy surge in literary brilliance, with numerous engrossing novels enthralling the hearts of readers worldwide. Lets delve into the realm of popular books, exploring the engaging narratives that have charmed audiences this year. The Must-Read : Colleen Hoover's "It Ends with Us" This touching tale of love, loss, and resilience has gripped readers with its raw and emotional exploration of domestic abuse. Hoover masterfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can triumph. Gas Law Simulation Lab Answer Key : Taylor Jenkins Reid's "The Seven Husbands of Evelyn Hugo" This captivating historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reid's captivating storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic : Delia Owens' "Where the Crawdads Sing" This evocative coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of captivating stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a brilliant and gripping novel that will keep you wondering until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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