Gas Steichkonsetey Practice 1.0 - Amorece May

Objective: - Perform stoichiometry calculations of gases including those at STP using molar volume

Directions: Solve each gas stochometry problem. Show all work. Include units.

 Calculate the volume of by drogen gas that well react with exactly 4.6 L of regigns gas using the equation below. Assume there is no change in temperature or pressure.

What volume of CO₂ gas will be produced from U.S.L. of oxygen gas reacting with excess others?

3. What volume of hydrogen at STP can be produced from the reaction of 36.1 g of your with hydrochiesis acid?

4. How many grams of NaCl can be produced by the reaction of 12 ft L of chlorine at STP with an excess of socioes?

 In order to produce 50.0 lines of animous gas (NOL) at NTP, how many grams of both hydrogen and satrogen would you need to react?

Gas Stoichiometry Practice Answer Key

Lauren Gardner

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