

- 1.) How many moles of gas occupy 98 L at a pressure of 2.8 atmospheres and a temperature of 292 K?
- 2.) If 5.0 moles of O_2 and 3.0 moles of N_2 are placed in a 30.0 L tank at a temperature of $25^\circ C$, what will the pressure of the resulting mixture of gases be?
- 3.) A balloon is filled with 35.0 L of helium in the morning when the temperature is $20.0^\circ C$. By noon the temperature has risen to $45.0^\circ C$. What is the new volume of the balloon?
- 4.) A 35 L tank of oxygen is at 315 K with an internal pressure of 190 atmospheres. How many moles of gas does the tank contain?

Gas Laws Worksheet Everett

Daniela Niemeyer

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