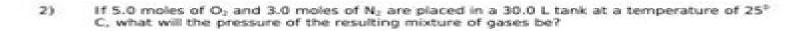




MIXED GAS LAWS WORKSHEET

1)	How many moles of gas occupy 98 L at a pressure of 2.8 atmospheres and a temperature of 292 K?



3) A balloon is filled with 35.0 L of helium in the morning when the temperature is 20.0° C. By noon the temperature has risen to 45.0° C. What is the new volume of the balloon?

4) A 35 L tank of oxygen is at 315 K with an internal pressure of 190 atmospheres. How many moles of gas does the tank contain?

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Daniela Niemeyer

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