

The Ideal and Combined Gas Laws

$$PV = nRT \quad \text{or} \quad \frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

Use your knowledge of the ideal and combined gas laws to solve the following problems. If it involves moles or grams, it must be $PV = nRT$

- 1) If four moles of a gas at a pressure of 5.4 atmospheres have a volume of 120 liters, what is the temperature?

$$(5.4)(120) = 4(.0821) T$$
$$T = 1973.2 \text{ K}$$

- 2) If I initially have a gas with a pressure of 84 kPa and a temperature of 35°C and I heat it an additional 230 degrees, what will the new pressure be? Assume the volume of the container is constant.

$$\frac{84}{(35+273)} = \frac{P_2}{(35+230+273)} \quad P_2 = 146.7 \text{ kPa}$$

- 3) My car has an internal volume of 2600 liters. If the sun heats my car from a temperature of 20°C to a temperature of 55°C , what will the pressure inside my car be? Assume the pressure was initially 760 mm Hg.

$$\frac{760 \text{ mmHg}}{(20+273)} = \frac{P_2}{(55+273)} \quad P_2 = 850 \text{ mmHg}$$

- 4) How many moles of gas are in my car in problem #3?

$$\frac{850 \text{ mmHg}}{760 \text{ mmHg}} = 1.13 \text{ atm}$$
$$2600(1.13) = n(.0821)(55+273)$$
$$n = 107.1 \text{ moles}$$

- 5) A toy balloon filled with air has an internal pressure of 1.25 atm and a volume of 2.50 L. If I take the balloon to the bottom of the ocean where the pressure is 95 atmospheres, what will the new volume of the balloon be? How many moles of gas does the balloon hold? (Assume $T = 285 \text{ K}$)

$$1.25(2.50) = 95(V_2)$$
$$V_2 = 0.032 \text{ L}$$
$$95(.032) = n(285)(.0821)$$
$$n = 0.1299 \text{ moles}$$

Gas Laws Test Answer Key

Jessica J Manson

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Gas Laws Test Answer Key:

This book delves into Gas Laws Test Answer Key. Gas Laws Test Answer Key is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Gas Laws Test Answer Key, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Gas Laws Test Answer Key
 - Chapter 2: Essential Elements of Gas Laws Test Answer Key
 - Chapter 3: Gas Laws Test Answer Key in Everyday Life
 - Chapter 4: Gas Laws Test Answer Key in Specific Contexts
 - Chapter 5: Conclusion
2. In chapter 1, this book will provide an overview of Gas Laws Test Answer Key. The first chapter will explore what Gas Laws Test Answer Key is, why Gas Laws Test Answer Key is vital, and how to effectively learn about Gas Laws Test Answer Key.
3. In chapter 2, the author will delve into the foundational concepts of Gas Laws Test Answer Key. The second chapter will elucidate the essential principles that need to be understood to grasp Gas Laws Test Answer Key in its entirety.
4. In chapter 3, the author will examine the practical applications of Gas Laws Test Answer Key in daily life. The third chapter will showcase real-world examples of how Gas Laws Test Answer Key can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Gas Laws Test Answer Key in specific contexts. The fourth chapter will explore how Gas Laws Test Answer Key is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Gas Laws Test Answer Key. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Gas Laws Test Answer Key.

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