

Question 1: Solutions

1. Define the following:
 - a. Intermolecular forces
 - b. Dispersion forces
 - c. Coulomb forces
 - d. Hydrogen

2. How does molecular structure affect boiling point? Give examples and explain. Then show boiling point trends for water, alcohols, acids, amines, and ethers. Then show boiling point trends for ionic compounds.

Boiling Point (°C)	Compound
100	H ₂ O
78	CH ₃ OH
60	CH ₃ CH ₂ OH
36	CH ₃ CH ₂ CH ₂ OH
15	CH ₃ CH ₂ CH ₂ CH ₂ OH

- a. Explain the boiling point trends.
- b. Explain the boiling point trends.
- c. Explain the boiling point trends.

Question 2: Polymers

- a. What is a polymer?
- b. What is a monomer?
- c. What is the difference between a polymer and a monomer?

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