

**Skills** – Use  $q = mc\Delta T$  to calculate the missing quantity, given the others.

Be sure to show all work required for the following calculations:

9. A 15.75g piece of iron absorbs 1006.75J of heat and its temperature changes from 25 to 175°C. Calculate the specific heat capacity of iron.

10. How many joules of heat are needed to raise the temp. of 10.0g of aluminum from 22 to 55°C, if the specific heat of aluminum is 0.90 J/g°C?

11. To what temperature will a 10.0g piece of glass raise if it absorbs 5275 J of heat and its specific heat capacity is 0.50 J/g°C? The initial temperature of the glass is 20°C.

12. Calculate the specific heat capacity of a piece of wood if 1500.0g of the wood absorbs 6.75x10<sup>5</sup>J, and its temp. changes from 32 to 57°C.

13. 100mL of 4.0°C water is heated until its temp. is 37°C. If the specific heat capacity of water is 4.18 J/g°C, calculate the amount of heat energy needed to cause this rise in temp. in joules AND calories.

14. 25.0g of an unknown substance is heated from 25 to 155°C, absorbing 455J in the process. Calculate the specific heat capacity of the unknown substance. Based on your answer determine the identity of the substance, using the chart on p329 of your text.

**Skills** – Calculate the enthalpy associated with a reaction.

15. Using the following balanced equation, calculate the thermal energy released ( $\Delta H$ ) when 45.0g of  $\text{NH}_4\text{NO}_3$  dissolved in water?



16. Calculate the change in enthalpy ( $\Delta H$ ) when 1.00g of hydrogen peroxide decomposes.



**Skills** – Calculate the energy of a reaction using Hess's Law and identify if it is exothermic or endothermic.

# Entropy Section Review Answers

**B Lingard**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## **Entropy Section Review Answers:**

Entropy Section Review Answers: Bestsellers in 2023 The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous compelling novels enthralling the hearts of readers worldwide. Lets delve into the realm of bestselling books, exploring the captivating narratives that have charmed audiences this year. The Must-Read : Colleen Hoover's "It Ends with Us" This heartfelt tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover skillfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can succeed. Entropy Section Review Answers : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This spellbinding historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids compelling storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Entropy Section Review Answers : Delia Owens "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These bestselling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of compelling stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a brilliant and gripping novel that will keep you guessing until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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