

## 2. - Elements and Compounds

1. Find chemical formulae for following groups compounds.

1(a) Sodium hydroxide

Answer:  $\text{NaOH}$   $\text{Na}$  is positive with the group

1(b) Calcium carbonate

Answer:  $\text{CaCO}_3$   $\text{Ca}$  is positive with the group

1(c) Aluminium

Answer:  $\text{Al}_2\text{O}_3$   $\text{Al}$  is positive

1(d) Magnesium sulphate

Answer:  $\text{MgSO}_4$   $\text{Mg}$  is positive with the group

1(e) Ammonia nitrate

Answer:  $\text{NH}_4\text{NO}_3$   $\text{N}$  is positive with the group

2. Find chemical formulae for

2(a) Silver chloride (silver has a positive charge of +1 and chlorine has a negative charge of -1)

Answer:  $\text{AgCl}$   $\text{Ag}$  is positive

2(b) Sodium hydroxide (sodium has a positive charge of +1 and hydroxide has a negative charge of -1)

Answer:  $\text{NaOH}$   $\text{Na}$  is positive

$\text{O}$  is negative

2(c) Nitric acid (nitrogen has a positive charge of +5 and oxygen has a negative charge of -2)

Answer:  $\text{HNO}_3$   $\text{N}$  is positive

2(d) Nitric acid (nitrogen has a positive charge of +5 and oxygen has a negative charge of -2)

Answer:  $\text{HNO}_3$   $\text{N}$  is positive

$\text{O}$  is negative

3. (a) Write down molecular formulae for the following compounds

3(a) Hydrogen

Answer:  $\text{H}_2$   $\text{H}$  is positive

3(b) Ethane (ethane has a positive charge of +2 and hydrogen has a negative charge of -1)

Answer:  $\text{C}_2\text{H}_6$   $\text{C}$  is positive

3(c)  $\text{CO}_2$  (carbon has a positive charge of +4 and oxygen has a negative charge of -2)

Answer:  $\text{CO}_2$   $\text{C}$  is positive

3(d)  $\text{H}_2\text{O}$  (hydrogen has a positive charge of +1 and oxygen has a negative charge of -2)

3(e) Ethane (ethane has a positive charge of +2 and hydrogen has a negative charge of -1)

Answer:  $\text{C}_2\text{H}_6$   $\text{C}$  is positive

$\text{H}$  is positive

3(f) Ethane (ethane has a positive charge of +2 and hydrogen has a negative charge of -1)

Answer:  $\text{C}_2\text{H}_6$   $\text{C}$  is positive

$\text{H}$  is positive

3(g)  $\text{CO}_2$  (carbon has a positive charge of +4 and oxygen has a negative charge of -2)

Answer:  $\text{CO}_2$   $\text{C}$  is positive

$\text{O}$  is negative

3(h) Ethane (ethane has a positive charge of +2 and hydrogen has a negative charge of -1)

Answer:  $\text{C}_2\text{H}_6$   $\text{C}$  is positive

$\text{H}$  is positive

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