Key

# Electron Configuration Evaluation

	ne space below, write the unabbreviated electron configurations of wring elements:	of the
1)	sodium /5 2 25 2 p 35 1	
2)	iron 152 252 2pt 352 3p6 4523	bd www. www. Wan
3)	bromine 152 252 2p6 3523p6 45230	110 4 20 30 11/1/16/1-1
4)	barium 152252p6 3523p6 4523d104p6	5524/1057 652
5)	neptunium 15 25 2p 435 3p 45 3 30 4p 55 4 40 10 51	106524F145410 BP
	he space below, write the abbreviated electron configurations of ments:	the following
6)	cobalt LAr 45 3d	
7)	silver	
8)	tellurium EKr 55° 401° 5p4	
9)	radium IRn 752	
10)	) lawrencium [Rn] 752 5F 14	
Det	termine what elements are denoted by the following electron cor-	nfigurations:
11)	) 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>4</sup>	
12)	1 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 4s <sup>2</sup> 3d <sup>10</sup> 4p <sup>6</sup> 5s <sup>1</sup> Rb	
13)	) [Kr] 5s <sup>2</sup> 4d <sup>10</sup> 5p <sup>3</sup> <u>A.5</u>	
14)	(Xe) 6s <sup>2</sup> 4f <sup>14</sup> 5d <sup>6</sup> 0.5	
15)	5) [Rn] 7s <sup>2</sup> 6d <sup>1</sup> 5f <sup>11</sup>	
The second	etermine which of the following electron configurations are not	did:
Del	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 4s (d <sup>1</sup> )p <sup>5</sup>	
2	2	
$\approx$		
(18	8) (Ra)7s <sup>2</sup> 5f <sup>8</sup>	
19		
20	o) [Xe]Xe	

# **Electron Configuration And Periodicity Review Answer Key**

**Mike Smith** 

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in Condensed Matter Physics where some say that rigor is probably impossible and certainly unnecessary By carefully select
ing the most important questions and formulating them as well defined mathemat ical problems and then solving a good
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