CHAPTER 8

CHEMICAL EQUILIBRIUM MCQs

| Q.1 | A reaction is reversible because | | | | |
|--------|---|---|-----------|---|--|
| | (a) | reactants are reactive | (b) | products are reactive | |
| | (c) | products are stable | (d) | reactants are stable | |
| Q.2 | A large value of Kc means that at equilibrium | | | | |
| | (a) | (a) less reactants and more products | | | |
| | (b) | more reactants and less product | | | |
| | (c) | same amount | 173 | | |
| | (d) | none | . 1 | | |
| Q.3 | Extent to H2 + I2 → 2HI can be increased by | | | | |
| | (a) | increasing pressure | (b) | increasing product | |
| | (c) | increasing temp. | Marian. | (d) adding a catalyst | |
| Q.4 | Strength of an acid can be determined by | | | | |
| | (a) | PKa | (b |) PKp | |
| | (c) | POH | (d) | PKw | |
| Q.5 | In an | exothermic reversible re | eaction i | increase in temp shifts the | |
| equili | ibrium t | to ot | | | |
| | (a) | reactant side | (b) | product side | |
| | (c) | remains unchanged | (d) | none | |
| Q.6 | Units of Kw are | | | | |
| | (a) | mole dm-3 | (b) | mole2 dm-3 | |
| | (c) | mole2 dm-6 | (d) | mole2 dm-3 | |
| Q.7 | A basic Buffer solution can be prepared by mixing | | | | |
| | (a) | | | | |
| | (b) | strong acid and its salt with weak base | | | |
| | (c) | weak base and its salt with strong acid | | | |
| | (d) | | | | |
| Q.8 | Buffer action can be explained by | | | | |
| | (a) | common ion effect | (b) | law of mass action | |

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