Question	Answer	Marks	Part marks
1(a)(i)	600 + (11+ 9) × 11 [=330] with no errors seen	MI	Could be in separate steps
I(a)(ii)	270	10	
1(b)(i)	372 cao nfww	3	B2 for answer 371.7 or M1 for 330 $\times \left(1 + \frac{1.5}{100}\right)^8$ oe not spoiled  After zero scored, SC1 for answer 42 or 41.7
l(b)(ii)	12.6 or 12.7 or 12.63 to 12.73	2	M1 for their (b)(i) -330 or their (b)(i) ×100 soi by 112.7 or 113 After zero scored, SC1 for answer 12%
1(c)(i)	99 280 cao final answer	1	
1(c)(ii)	27.5[0]	3	M2 for $24.75 + \frac{100-10}{100}$ oc or M1 for recognising $24.75$ as $90[\%]$ oc
1(d)(i)	32 cao	2	M1 for $\left(1 - \frac{20}{100}\right) \left(1 - \frac{15}{100}\right) [x]$ or for $0.15 \times 0.8 [x]$ oe
1(d)(ii)	13 сао	2	M1 for $\left(1 - \frac{20}{100}\right) \left(1 - \frac{15}{100}\right) \times x = 40.84 - 32$ oe seen or for their (d)(i) + $\left(1 - \left(\frac{their\ (d)(i)}{100}\right)\right) x = 40.84$ oe
2(a)(i)	Image at (8, 1), (10, 5), (8, 5)	2	<b>B1</b> for translation $\binom{6}{k}$ or $\binom{k}{-5}$ or 3 correct points not joined
2(a)(ii)	Image at (4, 10), (4, 8), (8, 8)	2	B1 for rotation 90° anticlockwise but different centre or for rotation 90° clockwise about (4, 10) or 3 correct points not joined
2(a)(iii)	Image at (6, 3), (6, 5), (7, 5)	2	B1 for enlargement factor $\frac{1}{2}$ but incorrect centre or 3 correct points not joined
2(b)	Reflection	1	
	y = -x oe	1	If zero scored, M1 for correct use of matrix product

# May 2013 Maths Igcse Mark Scheme 4h

SJ Ball

May 2013 Maths Igcse Mark Scheme 4h:

The Top Books of the Year May 2013 Maths Igcse Mark Scheme 4h The year 2023 has witnessed a noteworthy surge in literary brilliance, with numerous engrossing novels captivating the hearts of readers worldwide. Lets delve into the realm of top-selling books, exploring the captivating narratives that have charmed audiences this year. The Must-Read: Colleen Hoovers "It Ends with Us" This touching tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover skillfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can prevail. May 2013 Maths Igcse Mark Scheme 4h: Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This captivating historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids compelling storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. May 2013 Maths Igcse Mark Scheme 4h: Delia Owens "Where the Crawdads Sing" This mesmerizing coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens crafts a tale of resilience, survival, and the transformative power of nature, entrancing readers with its evocative prose and mesmerizing setting. These popular novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of engaging stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a exceptional and suspenseful novel that will keep you speculating until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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web 1 molarity mv mass molar mass x  $0.4500 \, l\, 0.825 \, g\, 141\, 9579 \, g\, mol \, x\, 0.0129 \, m\, 2$  molality  $0.825 \, g\, 141\, 9579 \, g\, mol \, 0.00581158 \, mol \, 0.00$ 

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web apr 7 2014 molarity moles of solute liters of solution 8 4 2 2 a first convert 250 ml to liters 250 1000 0 25 then calculate molarity 5 moles 0 25 liters 20 m 3 c a solution with molarity 2 requires 2 m of n a oh per liter so  $4 \times 2 \times 8 = 4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 8 = 4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 8 = 4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 8 = 4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 8 = 4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 8 = 4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 8 = 4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 8 = 4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 8 = 4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 8 = 4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 8 = 4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 8 = 4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 8 = 4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 10^{-2}$  m of n a oh per liter so  $4 \times 2 \times 10^{-2}$  m of n a oh per liter so  $4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 10^{-2}$  m of n a oh per liter so  $4 \times 10^{-2}$  m of n a oh per liter

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web answer the correct option is c explanation the molarity of a mixture mmix can be calculated using the following formula m mix m 1 v 1 m 2 v 2 v 1 v 2 1 5 x 480 1 2 x 520 480 520 720 624 1000 1 344 m q5 with increase in temperature which of these changes a molality b fraction of solute present in water

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web 1 calculate the molarity of a solution which contains 0 40 mol of a substance dissolved in 1 6 l of a solution 2 what is the molarity of a solution containing 325 g of nacl dissolved in 750 ml of solution 3 140 g of kcl is dissolved in 600 ml of water what is the molarity 4 724 4 g of ammonium phosphate in 4500 ml of alcohol

# 6 1 calculating molarity problems chemistry libretexts - Jun 13 2023

web determine the molarity for each of the following solutions 0 444 mol of cocl 2 in 0 654 l of solution 98 0 g of phosphoric acid h 3 po 4 in 1 00 l of solution 0 2074 g of calcium hydroxide ca oh 2 in 40 00 ml of solution 10 5 kg  $^{\circ}$ 

# 16 11 molality chemistry libretexts - Aug 15 2023

web molality differs from molarity only in the denominator while molarity is based on the liters of solution molality is based on the kilograms of solvent concentrations expressed in molality are used when studying properties of solutions related to vapor pressure and temperature changes

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