

1.2. Identify the products resulting from alpha decay of the following radioisotopes.

- | | |
|-----------------------------|----------------------------|
| 040 ^{238}Pu _____ | 075 Lanthanum-138 _____ |
| 040 ^{239}Po _____ | 080 Bismuth-213 _____ |
| 040 ^{238}Fr _____ | 090 Protactinium-233 _____ |
| 040 ^{238}Ra _____ | 095 Lanthanum-138 _____ |
| 040 ^{238}Ac _____ | 100 Uranium-235 _____ |

1.3. Identify the products resulting from beta decay of the following radioisotopes.

- | | |
|-----------------------------|-------------------------|
| 040 ^{238}Pa _____ | 075 Bismuth-213 _____ |
| 040 ^{238}Li _____ | 080 Calcium-45 _____ |
| 040 ^{238}F _____ | 090 Phosphorus-32 _____ |
| 040 ^{238}Sr _____ | 095 Carbon-14 _____ |
| 040 ^{238}Rb _____ | 100 Potassium-40 _____ |

1.4. Protactinium-233 has a half-life of 27.3 days.
How much of a 2 kg sample will remain after:

- | | |
|-----------------------|-------------------------|
| 040 4000 years? _____ | 040 14,000 years? _____ |
| 040 8000 years? _____ | 040 40,000 years? _____ |

1.5. What are three ways that nuclear reactions are different to chemical reactions?

- 03 _____
- 040 _____
- 040 _____

Key Terms Radioactivity And Nuclear Reactions Chapter 18 Answer

LL Leslie

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