

# AQA Chemistry

## GCSE Exam-style mark scheme

### C4 Chemical calculations

100 marks (20%)

1 hour

Question	Answer	Mark	Mark	Mark
10.1.1	14.2 g		1	100
10.1.2	14.2 g	<p>Amount of magnesium carbonate  <math>14.2 \text{ g} \div 142.2 \text{ g mol}^{-1}</math>  <math>= 0.1 \text{ mol}</math>  <math>\times 142.2 \text{ g mol}^{-1}</math>  <math>= 14.2 \text{ g}</math></p>	10	100
10.1.3	<p>any 1000 factor</p> <ul style="list-style-type: none"> <li>converting grams</li> <li>into kg (multiplication)</li> <li>needed</li> <li>into kg (multiplication) needed</li> <li>into the equation</li> </ul>		10	100
10.2	<p>all calculations must be            shown            amounts all used in the            calculation            all units must be correct            must be correct            must be correct            must be correct            must be correct</p>	<p>Amount of magnesium carbonate  <math>14.2 \text{ g} \div 142.2 \text{ g mol}^{-1}</math>  <math>= 0.1 \text{ mol}</math>  <math>\times 142.2 \text{ g mol}^{-1}</math>  <math>= 14.2 \text{ g}</math></p>	10	100
10.3.1	<p>all calculations must be            shown            amounts all used in the            calculation            all units must be correct            must be correct            must be correct            must be correct            must be correct</p>	<p>Amount of magnesium carbonate  <math>14.2 \text{ g} \div 142.2 \text{ g mol}^{-1}</math>  <math>= 0.1 \text{ mol}</math>  <math>\times 142.2 \text{ g mol}^{-1}</math>  <math>= 14.2 \text{ g}</math></p>	10	100
10.3.2	<p>all calculations must be            shown            amounts all used in the            calculation            all units must be correct            must be correct            must be correct            must be correct            must be correct</p>	<p>Amount of magnesium carbonate  <math>14.2 \text{ g} \div 142.2 \text{ g mol}^{-1}</math>  <math>= 0.1 \text{ mol}</math>  <math>\times 142.2 \text{ g mol}^{-1}</math>  <math>= 14.2 \text{ g}</math></p>	10	100
10.3.3	<p>all calculations must be            shown            amounts all used in the            calculation            all units must be correct            must be correct            must be correct            must be correct            must be correct</p>	<p>Amount of magnesium carbonate  <math>14.2 \text{ g} \div 142.2 \text{ g mol}^{-1}</math>  <math>= 0.1 \text{ mol}</math>  <math>\times 142.2 \text{ g mol}^{-1}</math>  <math>= 14.2 \text{ g}</math></p>	10	100

# Jun 2013 C4 Mark Scheme

**Guillaume Favre**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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