Nuclear Equations Worksheet

1. Complete the following nuclear reactions:

a)
$$^{10}_{17}C1 + ^{1}_{0}n \rightarrow ^{10}_{10}S +$$

d)
$$^{248}_{92}U + ^{12}_{6}C \rightarrow ^{246}_{98}Cf +$$

- Write the balanced equations for these nuclear reactions.
- a) Neutron emission by ^mBr
- b) Electron absorption by 116 Sb
- c) Positron emission by 184 Hg
- d) Alpha emission by 229 Th
- e) Neutron capture by 200 Hg
- 3. Match these statements with a given choice (A, B, or C). They may be used more than once.

- i. The type of reaction in the sun.
- ii. The type of reaction in atomic bombs.
- iii. An atomic nucleus is split into two roughly equal parts.
- iv. Requires incredibly high temperatures.
- v. Type of reaction in a nuclear power plant.
- vi. Takes place at room temperature.
- vii. 2 very light isotopes form a heavier one.
- viii. Main type of reaction in a hydrogen bomb.

<u>Introduction To Nuclear Chemistry Nuclear Equations</u> <u>Worksheet Answers</u>

Puja Mehta

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