#### Problem 1.6

Make a guess at the order of magnitude of the mass (e.g., 0.01, 0.1, 1.0, 10, 100, or 1000 lbm or kg) of standard air that is in a room 10 ft by 10 ft by 8 ft, and then compute this mass in lbm and kg to see how close your estimate was.

#### Solution

Given: Dimensions of a room.

Find: Mass of air in lbm and kg.

The data for standard air are:

$$R_{air} = 53.33 \cdot \frac{ft \cdot lbf}{lbm \cdot R}$$
  $p = 14.7 \cdot psi$   $T = (59 + 460) \cdot R = 519 \cdot R$ 

Then 
$$\rho = \frac{p}{R_{air} T}$$

$$\rho \ = \ 14.7 \cdot \frac{lbf}{in^2} \times \frac{1}{53.33} \cdot \frac{lbm \cdot R}{ft \cdot lbf} \times \frac{1}{519 \cdot R} \times \left(\frac{12 \cdot in}{1 \cdot ft}\right)^2$$

$$\rho = 0.0765 \frac{lbm}{ft^3}$$
 or  $\rho = 1.23 \frac{kg}{m^3}$ 

The volume of the room is 
$$V = 10$$
-ft ×  $10$ ft ×  $8$ ft  $V = 800$  ft<sup>3</sup>

The mass of air is then 
$$m = \rho \cdot V$$

$$m = 0.0765 \cdot \frac{1bm}{n^3} \times 800 \cdot ft^3$$
  $m = 61.21bm$   $m = 27.8 kg$ 

# **Introduction To Fluid Mechanics Solutions Manual**

**Camilla Rothe** 

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