

Model 2 – Prefixes and Suffixes

| Prefix | Numerical Value |
|--------|-----------------|
| mono- | 1 |
| di- | 2 |
| tri- | 3 |
| tetra- | 4 |
| penta- | 5 |
| hexa- | 6 |
| hepta- | 7 |
| octa- | 8 |
| nona- | 9 |
| deca- | 10 |

| Molecular Formula | Name of Compound |
|---------------------------|--------------------------|
| BCl_3 | Boron trichloride |
| SF_6 | Sulfur hexafluoride |
| IF_7 | Iodine heptafluoride |
| NI_3 | Nitrogen triiodide |
| N_2O_4 | Dinitrogen tetroxide |
| Cl_2O | Dichlorine monoxide |
| P_4O_{10} | Tetraphosphorus decoxide |
| B_5H_9 | Pentaboron nonahydride |
| Br_2O_8 | Tribromine octoxide |
| ClF | Chlorine monofluoride |

- Examine the prefixes in Model 2. Fill in the numerical value that corresponds to each prefix.
- What suffix (ending) do all the compound names in Model 2 have in common?

They all end in "ide".

- Carefully examine the names of the compounds in Model 2. When is a prefix NOT used in front of the name of an element?

*When the first element is one. (Examples: IF_7 - iodine heptafluoride)
 NI_3 - nitrogen triiodide)*

- Consider the compound NO.

- Which element, nitrogen or oxygen, would require a prefix in the molecule name? Explain your answer.

Oxygen would require a prefix of "mono" because it is the second element and there is one oxygen. Nitrogen does not require a prefix because it is the first element and there is one nitrogen.

- Name the molecule NO.

*Nitrogen monoxide (Notice, the "o" of mono was dropped)
 ↳ not monoxide*

- Find two compounds in Model 2 that contain a subscript of "4" in their molecular formula.

- List the formulas and names for the two compounds.

N_2O_4 - dinitrogen tetroxide || P_4O_{10} - Tetraphosphorus decoxide

- What is different about the spelling of the prefix meaning "four" in these two names?

The "a" has been dropped from the end of the prefix tetra.

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Shasha Hu



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