

Atomic Theory Webquest Part 1 History of the Discovery of the Atom

Name _____

In this activity you will follow the progression through time of how scientists used experimental evidence to refine our collective understanding of matter's essential building block: the atom.

Stage 1: "Discovering" the Atom: Democritus and Dalton

Use the following two links to read and learn about Democritus. Use the information in these links or other resources of your choosing to answer the questions below.

Link 1: http://www.abcte.org/files/previews/chemistry/s1_p1.html

Link 2: <https://www.sutori.com/en/story/democritus-XoJ4P9eA4pMMyxjx0ESZcis>

Crash Course: <https://www.youtube.com/watch?v=thnDxFdkezs>

1. Why was Democritus considered a philosopher and not a scientist?

2. When did Democritus develop his atomic theory?

3. What did the word "atom" originally mean in Greek? How did this relate to Democritus' atomic theory?

4. What are two ways in which Democritus' atomic theory differs from our understanding of atoms today? You may need to read more and come back to this question.

The next character in the atomic drama is John Dalton. Use these two links to learn more about Dalton.

Link 3: <https://www.differencebetween.com/difference-between-democritus-and-dalton-atomic-theory/>

Link 4: http://www.abcte.org/files/previews/chemistry/s1_p2.html

5. When did Dalton develop his atomic theory?

6. How much time passed between Democritus and Dalton?

History Of Atomic Theory Webquest Answers

Charles Daubeny

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From Atomics to Atom Andreas Gerardus Maria van Melsen,1960 *An Introduction to the Atomic Theory* Charles Daubeny,1851 **The Progress of Atomic Theory** Albert Cushing Crehore,1926 The Development of the Atomic Theory Andrew Norman Meldrum,2012-01 Unlike some other reproductions of classic texts 1 We have not used OCR Optical Character Recognition as this leads to bad quality books with introduced typos 2 In books where there are images such as portraits maps sketches etc We have endeavoured to keep the quality of these images so they represent accurately the original artefact Although occasionally there may be certain imperfections with these old texts we feel they deserve to be made available for future generations to enjoy John Dalton and the Atom Frank Greenaway,1966 Atomic Theory ,1927

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