

- (c) 350 cm, convergent  
(d) 150 cm, divergent
16. Three resistances A, B and C each of  $5\Omega$  are connected to a battery of 10 V as shown in the figure. Calculate the current through C:
- 
- (a) 2 amp (b) 1 amp  
(c) 3 amp (d) 0.6 amp
17. The device used to generate electrical energy is:
- (a) Electric motor (b) Generator  
(c) Galvanometer (d) Voltmeter
18. The number of molecules in 11 g of  $\text{CO}_2$  are:
- (a)  $0.25 \times 10^{23}$  (b)  $0.50 \times 10^{23}$   
(c)  $1.00 \times 10^{23}$  (d)  $1.51 \times 10^{23}$
19. The value of charge / mass ratio of electron was determined by:
- (a) W.K. Roentgen (b) J.J. Thomson  
(c) Marie Curie (d) Niels Bohr
20. Which of the following reactions is an example of combination reaction?
- (a)  $\text{Zn(s)} + \text{CuSO}_4(\text{aq}) \rightarrow \text{ZnSO}_4(\text{aq}) + \text{Cu(s)}$   
(b)  $\text{BaCl}_2(\text{aq}) + \text{Na}_2\text{SO}_4(\text{aq}) \rightarrow \text{BaSO}_4(\text{s}) + 2\text{NaCl}(\text{aq})$   
(c)  $\text{CaCO}_3(\text{s}) \rightarrow \text{CaO(s)} + \text{CO}_2(\text{g})$   
(d)  $\text{C(s)} + \text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g})$
21. The element 'X' and 'Y' have atomic numbers 12 and 17 respectively. Element 'X' reacts with element 'Y' to form compound with molecular formula:
- (a) XY (b)  $\text{XY}_2$   
(c)  $\text{XY}_3$  (d)  $\text{X}_3\text{Y}_3$
22. The atomic radius (pm) of Li, Na, K and Rb varies in the order:
- (a)  $\text{Na} < \text{K} < \text{Rb} < \text{Li}$  (b)  $\text{K} < \text{Na} < \text{Li} < \text{Rb}$   
(c)  $\text{Li} < \text{Na} < \text{K} < \text{Rb}$  (d)  $\text{Rb} < \text{K} < \text{Na} < \text{Li}$
23. The electron affinity value ( $\text{kJ mol}^{-1}$ ) of fluorine (F) is less than:
- (a) Hydrogen (H) (b) Lithium (Li)  
(c) Oxygen (O) (d) Chlorine (Cl)
24. Which of the following is an example of strong electrolyte?
- (a)  $\text{H}_2\text{CO}_3$  (b)  $\text{NH}_4\text{OH}$   
(c) NaCl (d)  $\text{COOH} - \text{COOH}$
25. If pH of a solution changes from 5 to 4, the change in hydrogen ion concentration shall be:
- (a) two times (b) five times  
(c) ten times (d) twenty times
26. Bleaching powder is manufactured by the reaction of:
- (a)  $\text{CaCl}_2$  and  $\text{CaCO}_3$  (b)  $\text{Cl}_2$  and  $\text{Ca(OH)}_2$
27. (c)  $\text{Cl}_2$  and  $\text{CaSO}_4$  (d)  $\text{Cl}_2$  and  $\text{Ca(HCO}_3)_2$   
The hard glass is obtained by fusing:
- (a) soda ash, sand and limestone  
(b) a mixture of sand, lime, borax and alkali carbonates  
(c) potassium carbonate and limestone  
(d) potassium carbonate, lead oxide and sand
28. Which of the following metals is most reactive?
- (a) Aluminium (b) lead  
(c) Mercury (d) Silver
29. Which of the following alloys contains chromium?
- (a) Steel (b) Stainless steel  
(c) Magnadium (d) Brass
30. Which of the following is monomer of natural rubber?
- (a) Chloroethene (b) Chloroprene  
(c) Isoprene (d) Buta-1, 3-diene
31. Alkaline  $\text{KMnO}_4$  oxidises propanone to:
- (a) propanoic acid (b) ethanoic acid  
(c) methanoic acid (d) oxalic acid
32. Which of the following metals can displace Zn from  $\text{ZnSO}_4$  solution?
- (a) Calcium (b) Copper  
(c) Iron (d) Mercury
33. Heating of sodium ethanoate with soda lime yields:
- (a) ethane (b) methane  
(c) ethanol (d) methanol
34. The compound formed by the reaction of ethyne with bromine is:
- (a)  $\text{Br}-\text{CH}=\text{CH}-\text{Br}$  (b)  $\text{Br}-\text{CH}_2-\text{CHBr}_2$   
(c)  $\text{CH}_2=\text{CH}-\text{Br}$  (d)  $\text{Br}_2\text{CH}-\text{CHBr}_2$
35. Cell were first discovered by Robert Hooke in the year:
- (a) 1665 (b) 1674 (c) 1831 (d) 1839
36. The plant cells have a rigid cell wall that lies:
- (a) Outside the plasma membrane  
(b) Inside the plasma membrane  
(c) In between the plasma membranes  
(d) None of the above
37. During mitosis nucleolus and nuclear membrane are lost in stage:
- (a) Prophase (b) Metaphase  
(c) Anaphase (d) Telophase
38. An example of simple and permanent tissue is:
- (a) Xylem (b) Phloem  
(c) Sclerenchyma (d) All the above
39. The shape of squamous epithelial tissue is:
- (a) Cubical (b) Flattened  
(c) Pillar like (d) None of the above
40. Fresh water sponge belongs to phylum:
- (a) Arthropoda (b) Annelida

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## **Table of Contents Entrance Exam Papers**

1. Understanding the eBook Entrance Exam Papers
  - The Rise of Digital Reading Entrance Exam Papers
  - Advantages of eBooks Over Traditional Books
2. Identifying Entrance Exam Papers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Entrance Exam Papers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Entrance Exam Papers
  - Personalized Recommendations
  - Entrance Exam Papers User Reviews and Ratings
  - Entrance Exam Papers and Bestseller Lists
5. Accessing Entrance Exam Papers Free and Paid eBooks
  - Entrance Exam Papers Public Domain eBooks
  - Entrance Exam Papers eBook Subscription Services
  - Entrance Exam Papers Budget-Friendly Options
6. Navigating Entrance Exam Papers eBook Formats
  - ePub, PDF, MOBI, and More
  - Entrance Exam Papers Compatibility with Devices
  - Entrance Exam Papers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Entrance Exam Papers
  - Highlighting and Note-Taking Entrance Exam Papers
  - Interactive Elements Entrance Exam Papers
8. Staying Engaged with Entrance Exam Papers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Entrance Exam Papers
- 9. Balancing eBooks and Physical Books Entrance Exam Papers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Entrance Exam Papers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Entrance Exam Papers
  - Setting Reading Goals Entrance Exam Papers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Entrance Exam Papers
  - Fact-Checking eBook Content of Entrance Exam Papers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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