

Element Builder

Answer Key

Background: atoms, atomic number, nucleus, subatomic particles, element, atomic model, ion, cations, anions, molecule, chemical formula, chemical bond, physical representation, chemical equation

Prior Knowledge Questions: (Choose ONE OF TWO using the items.)

Option 1: The proposed chemical equation is not well used, since subscripts and parentheses indicate **atoms** and **molecules** (not ions) are involved. The correct is $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ (hydrogen + oxygen).

Option 2: What is wrong with the different substances that make up water? (Choose 2 answers.)

(Protons, neutrons, electrons, nucleus, atoms, bonds, ions, molecules)

Option 3: What subatomic make up water? (Choose **subatomic** items.)

Option 4: What particles make up water? (Choose **atoms**.)

Answers: are not comprehensive that they make up water (lead to **atoms**). Protons had not electrons. Electrons is at the core of many molecules. What is a proton/electron. Both forming part. Some at atom, oxygen and hydrogen (not a an element because it is composed of two different atoms).

Class Warm-up

Atoms are tiny particles that are made up of three parts: **protons**, **neutrons**, and **electrons**. The **element** symbol for the atoms with four different kinds of protons. The particles inside form what gives atoms different **charges**.

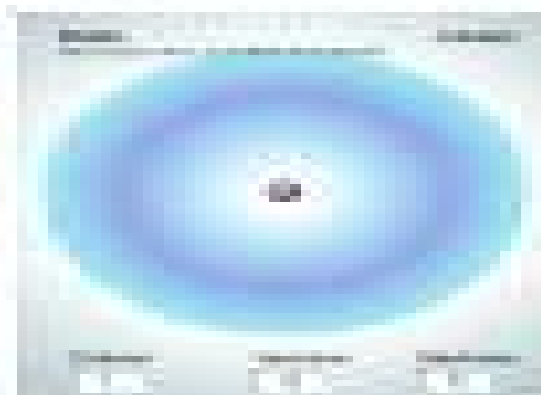
Option 1: Label the subatomic (+, -) to help students understand what electrons do in water. (Proton **Plus** (-))

Option 2: (Proton particles are located in the nucleus.)

(Protons **subatomic**.)

Option 3: Which particles will attract the nucleus?

(Electrons)



Option 4: How are **atoms** related to water? What causes the different kinds of charges?

(How changing **number** of protons gives a **hydrogen** atom a charge.)

Key For Elementbuilderse

C. V. G. Vallabhan, Mehmet Zülfü Aşık

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Key Elements - N, P, As, Sb, Bi, C-1 W. Pies, A. Weiss, 1978-09 The Third Element, 2008 **A Finite Element Method Primer for Mechanical Design** Charles E. Knight, 1994 *Finite Element Method for Engineers* C. V. G. Vallabhan, Mehmet Zülfü Aşık, 2011

Finite Element Method for Engineers From Theory to Practice while emphasizing problems in solid mechanics with numerous examples explains the theory in four fundamental steps with a flow chart to develop equations for the finite element models and the associated computer programs The theory is presented using energy principles in solid mechanics and towards the end the methodology is explained using the weak form of the Galerkin method so that the method can be used to solve flow problems and even some differential equations

KEY FEATURES With basic knowledge of strength of materials self learn the finite element method Simple subroutines given from which a beginner can write codes in a suitable computer language One can write own finite element computer codes with the information provided

Finite Element Methods Darrell W. Pepper, Juan C. Heinrich, 1992-02 Deals with the fundamentals of the finite element method Beginning with the concept of one dimensional heat transfer the book progresses through two dimensional elements and ultimately ends with a discussion on three dimensional elements Each chapter contains a set of example problems and exercises Overall the book is useful in describing how to develop and utilize finite element methodology to numerically solve problems

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