

- (c) Hydrochloric acid (d) Sulphuric acid
 (15) The alloy of copper (Cu) and Zinc (Zn) is:
 (a) Bronze (b) Brass (c) Nichrome (d) Amalgam
 (16) This gas is also called "Marsh gas".
 (a) Water gas (b) Mustard gas
 (c) Methane (d) Carbon monoxide
 (17) The chemical process for the preparation of soap is called:
 (a) Titration (b) Distillation
 (c) Neutralization (d) Saponification

CHEMISTRY

2014

Time: 2 ½ Hours

Max. Marks: 68

SECTION "B" (SHORT-ANSWER QUESTIONS)(42)

NOTE: Answer 14 questions from this section.

- (2) Write three sentences on the importance of Chemistry.
 (3) How many electrons, protons and neutrons are present in the following ions:
 (i) O^{2-} (ii) Mg^{+2} (iii) Cl^{-1}
 $[O^{16}_8, Mg^{24}_{12}, Cl^{35}_{17}]$
 (4) Calculate the number of atoms, present in one gram of Oxygen.
 (5) Balance the following chemical equations:
 (i) $S + H_2SO_4 \rightarrow SO_2 + H_2O$
 (ii) $C_2H_2 + O_2 \rightarrow CO_2 + H_2O$
 (iii) $NH_3 + Cl_2 \rightarrow N_2 + HCl$
 (6) Write the names of functional groups and formulae of the following:
 (i) Ethyl Alcohol (ii) Acetic Acid (iii) Propyl Iodide
 (7) Define double salt and write the formulae of:
 (i) Potash Alum (ii) Carnalite
 (8) Define: (i) a.m.u. (ii) mole (iii) mass number
 (9) Write 3 salient features of Mendeleev's periodic table.
 (10) Calculate molarity of solution containing 9.8 g H_2SO_4 per 500 mL solution.
 (11) Write any three characteristics of Transition elements.
 (12) Define: (i) Boiling Point (ii) Freezing Point (iii) Diffusion
 (13) Define: (i) Electro Chemistry Equivalent (ii) Electrolysis
 (14) Write three differences between diamond and graphite.
 (15) What is chemical formula of Bleaching Powder? Write its Laboratory Preparation.
 (16) Define: (i) Standard Solution
 (ii) Acidity of base (iii) Basicity of acid
 (17) Define refining of Petroleum and write two fractions of it.

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