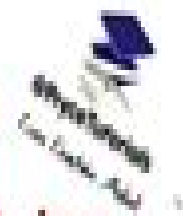


# Volumetric Analysis Calculations



30.0 cm<sup>3</sup> of 0.100 mol dm<sup>-3</sup> NaOH reacted completely with 25.0 cm<sup>3</sup> of H<sub>2</sub>SO<sub>4</sub> in a titration.

Calculate the concentration of H<sub>2</sub>SO<sub>4</sub> in mol dm<sup>-3</sup>?

Step 2: Find no' of moles in NaOH



✓  $c = 0.1 \text{ mol dm}^{-3}$

$V = 25 \text{ cm}^3$

✓  $V = \underline{30 \text{ cm}^3}$



Using  $n = c \times v$

$$\begin{aligned} n(\text{NaOH}) &= 0.1 \times \frac{30}{1000} \\ &= 0.003 \text{ mol} \end{aligned}$$

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# June 2015 Zimsec Chemistry Paper 2 O Level

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In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing short of extraordinary. Within the captivating pages of **June 2015 Zimsec Chemistry Paper 2 O Level** a literary masterpiece penned by way of a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of those that partake in its reading experience.

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