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Report post 1 Hour Ago 1

Post your answers below, and we'll get an unofficial mark scheme!



TheArson121

Thread Starter

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Random answer:

Bond energies - 1276

Explain why: Energy needed making bonds was greater than energy released breaking bonds

Balance equation: 2 and 2 somewhere

Efficiency of kettle: Hydrogencarbonate ions thermally decompose to carbonate, then react with calcium ions to produce a precipitate of calcium carbonate (limescale). Limescale is a bad conductor of electricity - more energy needed to heat water

Hard water cleaner i.e. Calcium and magnesium ions replaced by sodium and hydrogen ions

Benefits of hard water: contains calcium which strengthens teeth and bones

Element that has low boiling point: F or He

What happens in the mixture of gases? Ammonia is liquified, then cooled - Nitrogen and Hydrogen are recycled
Specific heat capacity: 3444J

How is methane obtained: Natural gas

Two boiling tubes: Tube A: fizzing because acid is dissolving calcium carbonate

Tube B: calcium carbonate produced, solution goes cloudy

Last edited by TheArson121, 42 minutes ago at 17:16



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