

(e) When reacted with sulfuric acid, lactic acid, C₃H₆O₃, loses water to form a compound with the molecular formula C₆H₈O₄. The infrared spectrum of this compound shows an absorption at 1735–1750 cm⁻¹ and no absorptions above 3000 cm⁻¹.

Suggest a structure for this compound, giving your reasoning.

- (f) When malic acid is heated, it loses water to form an acid anhydride with the molecular formula C₄H₂O₃.
 - Suggest a structure for this compound.

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