

# FORM THREE END TERM ONE EXAM 2020

## CHEMISTRY PAPER 1

Time: 1 Hour

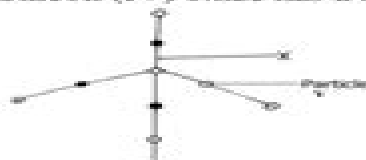
1. a) Distinguish between ionization energy and electron affinity. (2mks)
- .....
- .....
- b) The atomic number of A and B are 9 and 17 respectively. Compare the electron affinity of A and B. Explain. (1mk)
- .....
- .....

2. Use the reaction scheme below to answer the questions that follow.



- i) Draw the structure of alcohol X. (1mk)
- ii) Name process Y. (1mk)
- .....
- .....
- iii) Write the molecular formula of the 5<sup>th</sup> member in which propene belong. (1mk)

3. Silicon (IV) oxide has a structure similar to that of diamond. Part of the structure is shown below.



- a) What does x represent? (1mk)
- .....
- b) What type of structure is shown by the diagram? (1mk)
- .....
- c) Predict one physical property of silicon (IV) oxide and explain how it is related to its structure. (1mk)

# June 2012 66chemistry Past Paper

**J Dewey**



**June 2012 66chchemistry Past Paper:**

## Whispering the Secrets of Language: An Emotional Journey through **June 2012 66chchemistry Past Paper**

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