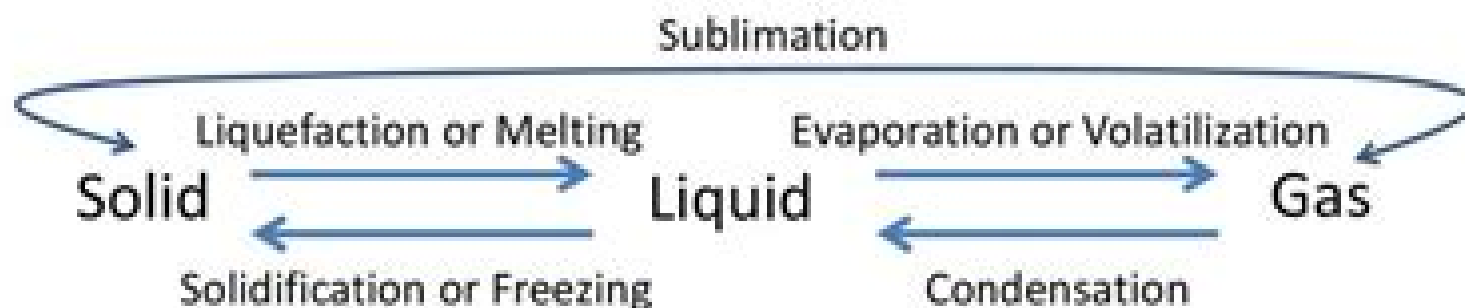


Phase Changes and Intermolecular Forces

- Melting points and boiling points can be used as indicators for intermolecular forces.
- For pressure of 1 bar, you have vapourization at the normal boiling point. Normal melting point is also at 1 bar pressure.
- Vapourization occurs when molecules escape the liquid phase more quickly than they are captured by the gas, reverse process is condensation.
- The stronger the intermolecular forces, the harder it is for the compound to melt or vapourize.



Investigation Question On Intermolecular Forces Of Evaporation

Christian G. Meyer



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