Problem 1.6

Make a guess at the order of magnitude of the mass (e.g., 0.01, 0.1, 1.0, 10, 100, or 1000 lbm or kg) of standard air that is in a room 10 ft by 10 ft by 8 ft, and then compute this mass in lbm and kg to see how close your estimate was.

Solution

Given: Dimensions of a room.

Find: Mass of air in lbm and kg.

The data for standard air are:

$$R_{air} = 53.33 \cdot \frac{ft \cdot lbf}{lbm \cdot R}$$
 $p = 14.7 \cdot psi$ $T = (59 + 460) \cdot R = 519 \cdot R$

Then $\rho = \frac{p}{R_{aie} T}$

$$\rho \ = \ 14.7 \cdot \frac{lbf}{in^2} \times \frac{1}{53.33} \cdot \frac{lbm \cdot R}{ft \cdot lbf} \times \frac{1}{519 \cdot R} \times \left(\frac{12 \cdot in}{1 \cdot ft}\right)^2$$

$$\rho = 0.0765 \frac{lbm}{ft^3}$$
 or $\rho = 1.23 \frac{kg}{m^3}$

The volume of the room is V = 10-ft × 10ft × 8ft V = 800 ft³

The mass of air is then $m = \rho \cdot V$

$$m = 0.0765 \cdot \frac{1bm}{n^3} \times 800 \cdot ft^3$$
 $m = 61.21bm$ $m = 27.8 kg$

Introduction To Fluid Mechanics Solution Manual

Donald F. Young, Bruce R.
Munson, Theodore H. Okiishi, Wade W.
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