# WORD EQUATIONS

Name <u>Key</u>

write the world equations below as chemical equations and balance.

1. zinc + lead (ii) nitrate yield zinc nitrate + lead

 $Z_{1} + P6(NO_{3})_{2} \rightarrow Z_{1}(NO_{3})_{2} + P6$ 

2. aluminum bromida + chlorine yield aluminum chloride + bromine

2A1Br, +3C/2 →2A1C/3+3Br2

 sodium phosphate + coloium chloride yield calcium phosphate + sodium chloride

 $ZNL_{1}, PO_{V(ng)}+3CaCl_{2(ng)} \longrightarrow Ca_{1}(PO_{y})_{2(n)}+6NhCl_{ng}$ 

4. potassium chlorate when heated yields potassium chloride + coygen gas

2 KC103 - →2KC1 +302

5. aluminum + hydrochloric acid yield aluminum chloride + hydrogen gas

2A1 +6HC1 - 2A1C1 +3H2

6. calcium hydroxide + phasphoric acid yield calcium phasphate + water

 $3 Ca(OH)_{2(m)} + 2H_{2}PO_{4m} \rightarrow Ca_{3}(PO_{6})_{2m} + 6H_{2}O_{6}$  $5H_{2}A+Y$  Soluble

7. copper + sulfuric acid yield copper (ii) sulfate + water + sulfur dioxide

Cu + 2H250y -> Cu50y +2H20 + 502

hydrogen + nitrogen monoxide yield water + nitrogen

 $2H_2 \rightarrow 2NO \rightarrow 2H_2O + N_3$ 

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