

ChemThink: Behavior of Gases

Name _____ Per _____

1. What does the speed of atoms depend on? Describe the relationship between speed and each of these factors.

temperature: \uparrow temp = \uparrow speed

type of atom: smaller/lighter = faster

2. In a sample of one type of atom, do all the atoms move at the same speed? Why?

No, speed is related to each individual atoms energy

3. What is the equation for Pressure and what are the units for pressure?

$$\text{Pressure} = \frac{\text{Force}}{\text{Area}} \quad \text{psi}$$

4. When atoms are moving in a closed container what is creating the force on the sides of the container?

impacts (collisions) of atoms w/ sides of container

5. Explain how a change in temperature causes pressure to change and include an equation.

more collisions results in higher pressure

\downarrow
b/c molecules moving faster

$$\frac{P}{T} = \text{constant}$$

6. Draw the graph you observed of pressure vs. temperature. Is this a direct relationship or an inverse relationship?



direct

7. Fill the cause/effect relationship in the diagram of boxes you observed.



- Explain how the number of atoms present causes pressure to change and include an equation?

more atoms = more collisions, which increases pressure

$$\frac{P}{n} = \text{constant}$$

How To Do Behavior Of Gases Chem Think

William Popenorf



How To Do Behavior Of Gases Chem Think:

Introduction to Chemical Structure Donald Cotter,2020-08-01 **School Science and Mathematics** ,1918

Chemistry in the Community (ChemCom) American Chemical Society,2011-06-17 Touted as the most successful NSF funded project published Chemistry in the Community ChemCom by the American Chemical Society ACS offers a meaningful and memorable chemistry program for all levels of high school students ChemCom covers traditional chemistry topics within the context of societal issues and real world scenarios Centered on decision making activities where students are responsible for generating data in an investigating analyzing that data and then applying their chemistry knowledge to solve the presented problem The text is intensively laboratory based with all 39 of the investigations integrated within the text not separate from the reading With the ChemCom program students learn more organic and biochemistry more environmental and industrial chemistry and more on the particulate nature of matter than other textbooks all within the relevance of solving problems that arise in everyday life Meticulously updated to meet the needs of today s teachers and students the new sixth edition of ChemCom adheres to the new science framework as well as the forthcoming next generation of science standards Incorporating advances in learning and cognitive sciences ChemCom s wide ranging coverage builds upon the concepts and principles found in the National Science Education Standards Correlations are available showing how closely aligned ChemCom is to these and other state standards **Chemistry** Carla Mooney,2016-05-16 Have you ever wondered what makes up everything in the world around you Or what exactly is the difference between solids liquids and gases Have you wanted to know what causes two substances to react or change Chemistry Investigate the Matter that Makes Up Your World introduces readers 12 through 15 to the fascinating world of protons neutrons and electrons Learn how these molecules combine to form ordinary objects such as the chair you re sitting on the water in your glass even you Through hands on investigative projects readers delve into the world of chemical reactions and changing matter learning how these principles are used in many areas of science from biochemistry to nuclear science Combining hands on science inquiry with chemistry mathematics and biology projects include building models of molecules and bonds identifying acids and bases investigating the effect of temperature on reaction rate and observing how a chemical reaction from vinegar water and bleach can accelerate the rusting of steel Chemistry offers entertaining illustrations and fascinating sidebars to illuminate the topic and engage readers further plus integrates a digital learning component by providing links to primary sources videos and other relevant websites *From Elements to Atoms* Robert Siegfried,2002 Seeking to enlarge an understanding of the nature of chemical science references to recently published secondary works are intended for the benefit of readers who wish to enlarge their perspectives on the development of early chemical thinking More Chemistry Basics William C. Robertson,2010 Overwhelmed by orbitals Terrified of thermodynamics Agitated by acids and bases Have no fear This follow up to the award winning Chemistry Basics will clear up your chemistry woes In More Chemistry Basics the ninth book in the

Stop Faking It series author Bill Robertson introduces additional chemistry concepts such as special reactions and half lives and expands on many previously discussed ideas including electron energy levels and why we can't know exactly what electrons are doing and where they are. Robertson explains science basics using easy to follow activities that help teachers learn the fundamentals and more. Like other books in this series More Chemistry Basics will prove invaluable for teachers, parents and home school providers who want to feel greater confidence in the content they teach. Robertson's humorous and honest approach will help you brush up on chemistry concepts and feel more prepared even excited to teach chemistry to your students.

An Introduction to Chemistry Michael Mosher, Paul Kelter, 2023-03-18 This textbook is written to thoroughly cover the topic of introductory chemistry in detail with specific references to examples of topics in common or everyday life. It provides a major overview of topics typically found in first year chemistry courses in the USA. The textbook is written in a conversational question based format with a well defined problem solving strategy and presented in a way to encourage readers to think like a chemist and to think outside of the box. Numerous examples are presented in every chapter to aid students and provide helpful self learning tools. The topics are arranged throughout the textbook in a traditional approach to the subject with the primary audience being undergraduate students and advanced high school students of chemistry.

Articulating the World Joseph Rouse, 2015-11-13 Naturalism both disavows any appeal to the supernatural or anything else transcendent to nature and repudiates any philosophical or religious authority over the workings and conclusions of the sciences. Paradoxically however scientific knowledge itself appears to transcend nature seemingly making it impossible to conceptualize within scientific naturalism. In *Articulating the World* Joseph Rouse takes up this challenge drawing on recent developments in evolutionary biology and the philosophy of science to defend naturalism by revising both how we understand our scientific conception of the world and how we situate ourselves within it.

Chemistry in the Community. American Chemical Society, 2002 This volume has relevance to a wide number of courses giving a hands on introduction to chemistry in relation to community issues rather than around specific chemical concepts.

American Gas-light Journal and Chemical Repertory, 1909 **The Freudian Paradigm** Mohammed Mujeeb-ur-Rahman, 1977 **Science**, 2009-10-05

The complete illustrated science encyclopedia covering the history, key discoveries, inventions and people. This remarkable reference book reveals the story of scientific progress from the invention of the wheel to 21st century climate solutions including everything from ancient Greek geometry and quantum physics to the worldwide web. Explore every key moment of scientific discovery and find out how the concepts, inventions and the individuals behind them have changed our world. With stunning artworks and authoritative information, this makes even complex scientific subjects easily comprehensible.

Student's Guide to Masterton and Slowinski's Chemical Principles Ray Boyington, William L. Masterton, 1977

Chemistry in the Community (Enhanced Core Four) American Chemical Society, 2006-02-15 **Industrial Hygiene Control of Airborne Chemical Hazards, Second Edition** William Popen Dorf, 2019-06-26 Are you a practicing

occupational hygienist wondering how to find a substitute organic solvent that is safer to use than the hazardous one your company is using Chapter 6 is your resource Are you a new hygienist looking for an alternative technology as a nonventilation substitute for an existing hazard Chapter 8 is your resource Are you looking for an overview of ventilation Chapters 10 and 11 are your resource Are you an industrial hygiene student wanting to learn about local exhaust ventilation Chapters 13 through 16 are your resource Are you needing to learn about personal protective equipment and respirators Chapters 21 and 22 are your resources This new edition brings all of these topics and more right up to date with new material in each chapter including new governmental regulations While many of the controls of airborne hazards have their origins in engineering this author has been diligent in explaining concepts writing equations in understandable terms and covering the topics of non ventilation controls both local exhaust and general ventilation and receiver controls at the level needed by most IHs without getting too advanced Taken as a whole this book provides a unique comprehensive tool to learn the challenging yet rewarding role that industrial hygiene can play in controlling airborne chemical hazards at work Most chapters contain a set of practice problems with the solutions available to instructors Features Written for the novice industrial hygienist but useful to prepare for ABIH certification Explains engineering concepts but requires no prior engineering background Includes specific learning goals that differentiate the depth of learning appropriate to each topic within the fuller information and explanations provided for each chapter Contains updated governmental regulations and abundant references Presents a consistent teaching philosophy and approach throughout the book Deals with both ventilation and non ventilation controls

Physical and Chemical Mechanisms in Molecular Radiation Biology William A. Glass, Matesh N. Varma, 2012-12-06 The fundamental understanding of the production of biological effects by ionizing radiation may well be one of the most important scientific objectives of mankind such understanding could lead to the effective and safe utilization of the nuclear energy option In addition this knowledge will be of immense value in such diverse fields as radiation therapy and diagnosis and in the space program To achieve the above stated objective the U S Department of Energy DOE and its predecessors embarked upon a fundamental interdisciplinary research program some 35 years ago A critical component of this program is the Radiological and Chemical Physics Program RCPP When the RCPP was established there was very little basic knowledge in the fields of physics chemistry and biology that could be directly applied to understanding the effects of radiation on biological systems Progress of the RCPP program in its first 15 years was documented in the proceedings of a conference held at Airlie Virginia in 1972 At this conference it was clear that considerable progress had been made in research on the physical and chemical processes in well characterized systems that could be used to understand biological effects During this period of time most physical knowledge was obtained for the gas phase because the technology and instrumentation had not progressed to the point that measurements could be made in liquids more characteristic of biological materials

Chemistry John S. Phillips, Cheryl Wistrom, 2000 *Resources in*

Education ,1990 Resources for Teaching Elementary School Science National Science Resources Center of the National Academy of Sciences and the Smithsonian Institution,1996-04-28 What activities might a teacher use to help children explore the life cycle of butterflies What does a science teacher need to conduct a leaf safari for students Where can children safely enjoy hands on experience with life in an estuary Selecting resources to teach elementary school science can be confusing and difficult but few decisions have greater impact on the effectiveness of science teaching Educators will find a wealth of information and expert guidance to meet this need in Resources for Teaching Elementary School Science A completely revised edition of the best selling resource guide Science for Children Resources for Teachers this new book is an annotated guide to hands on inquiry centered curriculum materials and sources of help in teaching science from kindergarten through sixth grade Companion volumes for middle and high school are planned The guide annotates about 350 curriculum packages describing the activities involved and what students learn Each annotation lists recommended grade levels accompanying materials and kits or suggested equipment and ordering information These 400 entries were reviewed by both educators and scientists to ensure that they are accurate and current and offer students the opportunity to Ask questions and find their own answers Experiment productively Develop patience persistence and confidence in their own ability to solve real problems The entries in the curriculum section are grouped by scientific area Life Science Earth Science Physical Science and Multidisciplinary and Applied Science and by type core materials supplementary materials and science activity books Additionally a section of references for teachers provides annotated listings of books about science and teaching directories and guides to science trade books and magazines that will help teachers enhance their students science education Resources for Teaching Elementary School Science also lists by region and state about 600 science centers museums and zoos where teachers can take students for interactive science experiences Annotations highlight almost 300 facilities that make significant efforts to help teachers Another section describes more than 100 organizations from which teachers can obtain more resources And a section on publishers and suppliers give names and addresses of sources for materials The guide will be invaluable to teachers principals administrators teacher trainers science curriculum specialists and advocates of hands on science teaching and it will be of interest to parent teacher organizations and parents The Chemistry Leaflet ,1928

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