

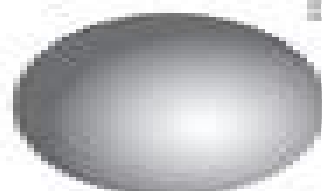
Name _____

Class _____

Date _____

Identify which atomic model is shown in each picture and write that name below the picture. Write the scientist responsible for the model below each picture as well. There is a word bank below to help you.

History of the Atom

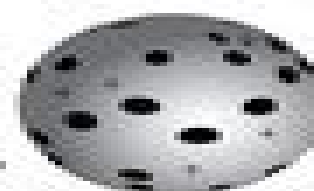


Indivisible atom

1a. Model _____

b. Scientist(s) responsible _____

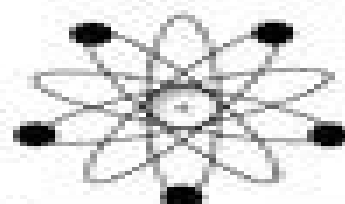
Scattered negatively charged small particles within the body



positively charged, low density, spacious body

2a. Model _____

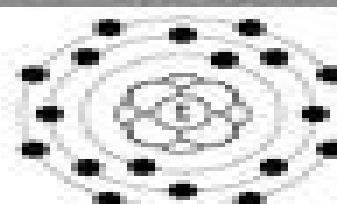
b. Scientist(s) responsible _____



positively charged center with orbiting negatively charged particles

3a. Model _____

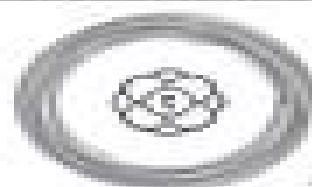
b. Scientist(s) responsible _____



Nucleus with protons, neutrons, and electrons must orbit at defined energy levels and electrons away from the nucleus

4a. Model _____

b. Scientist(s) responsible _____



Nucleus with electrons that have calculated probable areas where they are located, but no exact location can be determined

5a. Model _____

b. Scientist(s) responsible _____

6. Now measure the following statement's from Dalton's Atomic Theory. Circle true or false for each statement and if it's false, write below what you know about the atom today that contradicts his theory

A. All matter is made of atoms.
True or False _____

B. Atoms are indivisible and indestructible.
True or False _____

C. All atoms of a given element are identical in mass and properties.
True or False _____

D. Compounds are formed by a combination of two or more different kinds of atoms.
True or False _____

Models that is chronological order 2
 primitive model
 simple, first atomic theory
 plum pudding model
 Bohr model
 Electron Cloud model

Scientists that is order 2
 Ernest Rutherford
 J.J. Thomson
 John Dalton
 Modern Scientists
 Niels Bohr

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Martin Shields



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