

Griddable Science Items

Topography map

Density

Atoms

$$F = ma$$

			6	.	2	5
0	0	0	0		0	0
1	1	1	1		1	1
2	2	2	2		2	2
3	3	3	3		3	3
4	4	4	4		4	4
5	5	5	5		5	5
6	6	6	6		6	6
7	7	7	7		7	7
8	8	8	8		8	8
9	9	9	9		9	9

Designed for

8th grade STAAR Test



but would work for other standardized tests

Crackdoables

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1. A balance and a graduated cylinder are used to determine the density of a mineral sample. The sample has a mass of 75.1 g and a volume of 14.2 cm³. What is the density of the mineral sample in g/cm³?

$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{75.1}{14.2} = 5.3 \text{ g/cm}^3$$

2. 1 mL = 1 cm³

What is the density of the metal in g/cm³?

$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{100}{10} = 10 \text{ g/cm}^3$$

3. The sample of metal has a mass of 25 g. What is the density in g/cm³?

$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{25}{50} = 0.5 \text{ g/cm}^3$$

4. A balance and a graduated cylinder are used to determine the density of a mineral sample. The sample has a mass of 75.1 g and a volume of 14.2 cm³. What is the density of the mineral sample in g/cm³?

$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{75.1}{14.2} = 5.3 \text{ g/cm}^3$$

5. 1 mL = 1 cm³

What is the density of the metal in g/cm³?

$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{100}{10} = 10 \text{ g/cm}^3$$

6. The sample of metal has a mass of 25 g. What is the density in g/cm³?

$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{25}{50} = 0.5 \text{ g/cm}^3$$

Answer each question about #1-6 in the problems, for each to use the

1.

2.

3.

4.

5.

6.

7.

8.

9.

Design

81

ST

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LL Leslie



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