

82. Additional Questions

1. The percent by mass composition of a salt was found to be 76.58% potassium, 8.48% sulfur, and 14.77% oxygen. What is the empirical formula of this salt?

$$56.58g \text{ of K} \left(\frac{\text{molar K}}{39.09g \cdot \text{mol}^{-1}} \right) = 1.447 \text{molar K} \quad \frac{1.447 \text{molar K}}{0.723 \text{molar S}} = 2$$

$$8.48g \text{ of S} \left(\frac{\text{molar S}}{32.06g \cdot \text{mol}^{-1}} \right) = 0.264 \text{molar S} \quad \frac{0.264 \text{molar S}}{0.264 \text{molar S}} = 1$$

$$14.77g \text{ of O} \left(\frac{\text{molar O}}{16.00g \cdot \text{mol}^{-1}} \right) = 0.923 \text{molar O} \quad \frac{0.923 \text{molar O}}{0.264 \text{molar S}} = 3$$

$$\text{empirical formula} = \text{K}_2\text{S}_2\text{O}_6$$

2. A compound consisting iron and sulfur was formed by combining 3.235 g of iron with 1.426 g of sulfur. What is the empirical formula of the compound?

$$3.235g \text{ of Fe} \left(\frac{\text{molar Fe}}{55.85g \cdot \text{mol}^{-1}} \right) = 0.05793 \text{molar Fe} \quad \frac{0.05793 \text{molar Fe}}{0.05793 \text{molar Fe}} = 1 + 2 = 2$$

$$1.426g \text{ of S} \left(\frac{\text{molar S}}{32.07g \cdot \text{mol}^{-1}} \right) = 0.04447 \text{molar S} \quad \frac{0.04447 \text{molar S}}{0.05793 \text{molar Fe}} = 0.7675 + 2 = 3$$

$$\text{empirical formula} = \text{Fe}_2\text{S}_3$$

3. Propylene has a molar mass of 42.08 g/mol and is composed of 84.3% hydrogen and 85.7% carbon. What is the molecular formula of propylene?

$$14.3g \text{ of H} \left(\frac{\text{molar H}}{1.01g \cdot \text{mol}^{-1}} \right) = 14.158 \text{molar H} \quad \frac{14.158 \text{molar H}}{7.129 \text{molar C}} = 2$$

$$85.7g \text{ of C} \left(\frac{\text{molar C}}{12.01g \cdot \text{mol}^{-1}} \right) = 7.129 \text{molar C} \quad \frac{7.129 \text{molar C}}{7.129 \text{molar C}} = 1$$

$$\text{empirical formula} = \text{CH}_2$$

$$\text{CH}_2 = 12.01 + 2(1.01) = 14.03 \quad \frac{42.08}{14.03} = 3$$

$$\text{molecular formula} = \text{C}_3\text{H}_6$$

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L Towne

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'L Towne' text.

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Table of Contents Emirical Formula Page 55 Answers

1. Understanding the eBook Emirical Formula Page 55 Answers
 - The Rise of Digital Reading Emirical Formula Page 55 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Emirical Formula Page 55 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Emirical Formula Page 55 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Emirical Formula Page 55 Answers
 - Personalized Recommendations
 - Emirical Formula Page 55 Answers User Reviews and Ratings
 - Emirical Formula Page 55 Answers and Bestseller Lists

5. Accessing Emirical Formula Page 55 Answers Free and Paid eBooks
 - Emirical Formula Page 55 Answers Public Domain eBooks
 - Emirical Formula Page 55 Answers eBook Subscription Services
 - Emirical Formula Page 55 Answers Budget-Friendly Options
6. Navigating Emirical Formula Page 55 Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Emirical Formula Page 55 Answers Compatibility with Devices
 - Emirical Formula Page 55 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Emirical Formula Page 55 Answers
 - Highlighting and Note-Taking Emirical Formula Page 55 Answers
 - Interactive Elements Emirical Formula Page 55 Answers
8. Staying Engaged with Emirical Formula Page 55 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Emirical Formula Page 55 Answers
9. Balancing eBooks and Physical Books Emirical Formula Page 55 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Emirical Formula Page 55 Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Emirical Formula Page 55 Answers
 - Setting Reading Goals Emirical Formula Page 55 Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Emirical Formula Page 55 Answers
 - Fact-Checking eBook Content of Emirical Formula Page 55 Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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