

4. It is not safe to put aerosol canisters in a campfire, because the pressure inside the canisters gets very high and they can explode. If I have a 1.0 liter canister that holds 2 moles of gas, and the campfire temperature is 1400°C , what is the pressure inside the canister?

What equation(s) should be used?

Solve:

5. If I place 3 moles of N_2 and 4 moles of O_2 in a 55 L container at a temperature of 25°C , what will the pressure of the resulting mixture of gases be?

What equation(s) should be used?

Solve:

6. At a pressure of 1.0 atm and a temp of 298K , calcium carbonate decomposes to form carbon dioxide and calcium oxide. How many grams of calcium carbonate will be needed to form 3.45 liters of carbon dioxide?

Review (Skill - Student will apply Kinetic Molecular Theory when describing a novel, real-world situation in terms of collisions and motions of particles)

7. A warmed empty bottle has a balloon attached to its top, as done in station 1 of Expt 40.00204. When the bottle is placed into a cold water bath, the balloon deflates & is pulled in to the bottle. Using the Kinetic Molecular theory, explain why this occurs, making sure to refer to the motion of the gas molecules.

Gases Practice Assessment:

Name _____

Modern Chemistry Review Gases Section 5 Answers

Neil D. Jespersen, Pamela Kerrigan



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Solutions Guide for Introductory Chemistry James F. Hall, 1996

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