

## Chapter 9 Carbon Chemistry

**Section 9.2 Substituted Hydrocarbons****(pages 272–274)**

*This section discusses organic compounds that contain atoms of elements other than carbon and hydrogen. It also explains the relationship between the properties of organic compounds and functional groups.*

**Reading Strategy (page 272)**

**Monitoring Your Understanding** As you read, complete the table by connecting each functional group with the type of compound that contains the functional group. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

| Connecting Functional Groups to Types of Compounds |                  |
|----------------------------------------------------|------------------|
| Functional Group                                   | Type of Compound |
| –OH                                                |                  |
| –COOH                                              |                  |
| –NH <sub>2</sub>                                   |                  |

1. Name the two main products when methane and chlorine react.
  - a. \_\_\_\_\_
  - b. \_\_\_\_\_
2. To which environmental problem have researchers connected halocarbons containing chlorine and fluorine? \_\_\_\_\_  
\_\_\_\_\_  
\_\_\_\_\_
3. Describe a substituted hydrocarbon. \_\_\_\_\_  
\_\_\_\_\_  
\_\_\_\_\_
4. Is the following sentence true or false? The functional group in a substituted hydrocarbon determines the properties of the compound. \_\_\_\_\_

**Alcohols (page 273)**

5. Methanol and ethanol are two examples of a class of organic compounds called \_\_\_\_\_.
6. The functional group in an alcohol is represented as –OH and is called a(n) \_\_\_\_\_ group.
7. Identify two ways a halocarbon can be produced.
  - a. \_\_\_\_\_
  - b. \_\_\_\_\_

# Hydrocarbon Section Review Answers

**B Lingard**



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